



# Greenhouse Gases Mitigation CO<sub>2</sub> Capture and Utilization

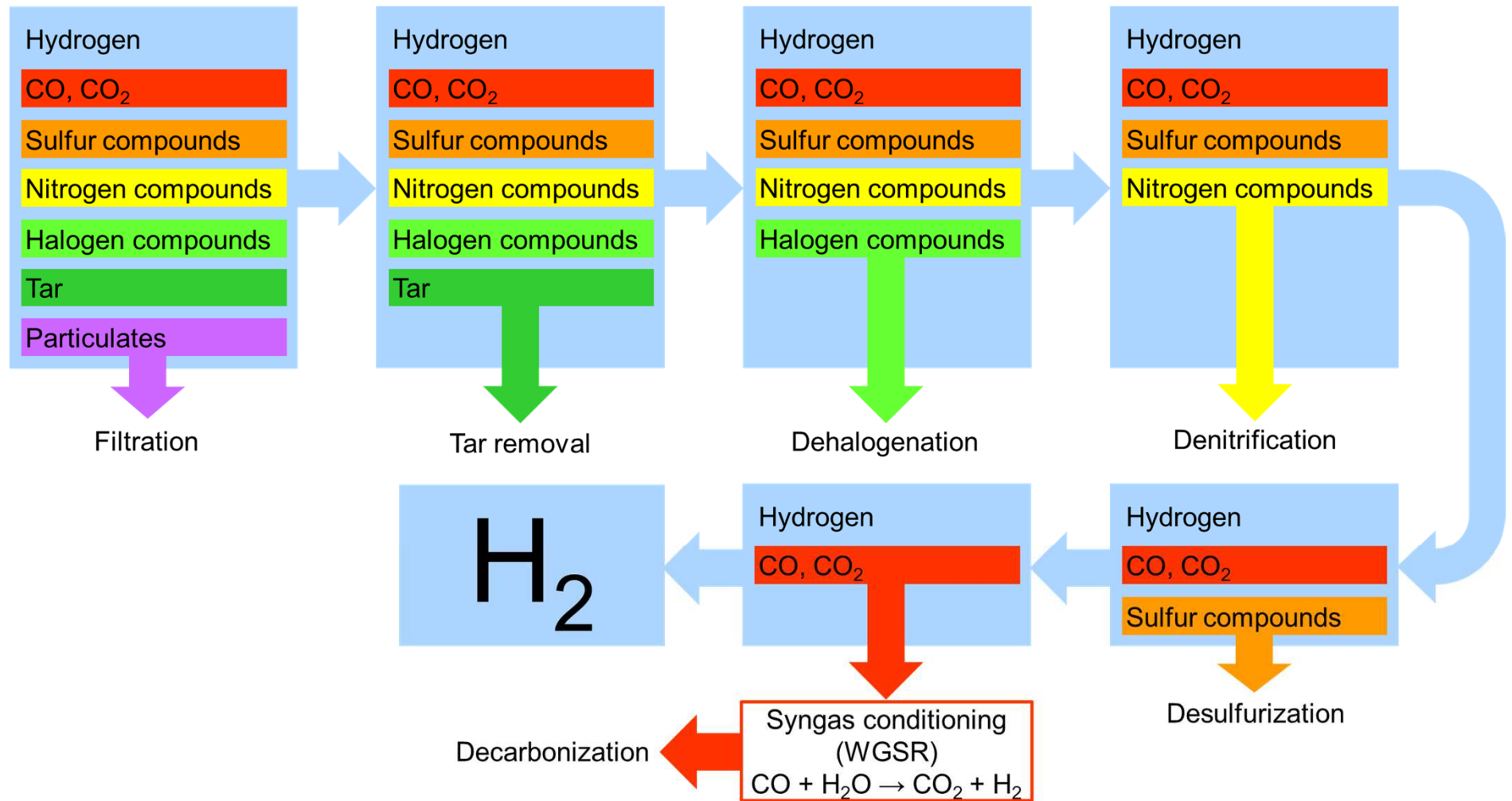
Topic No: 4



1. Pre-combustion processes: syngas contaminants removal
2. Post-combustion processes: flue gas contaminants removal
3. Overview of methods available for CO<sub>2</sub> capture from syngas and flue gas
4. General scheme of absorption CO<sub>2</sub> removal processes
5. General scheme of pressure swing adsorption
6. General scheme of high temperature carbonate looping



- A typical sequence of syngas cleaning and conditioning steps:





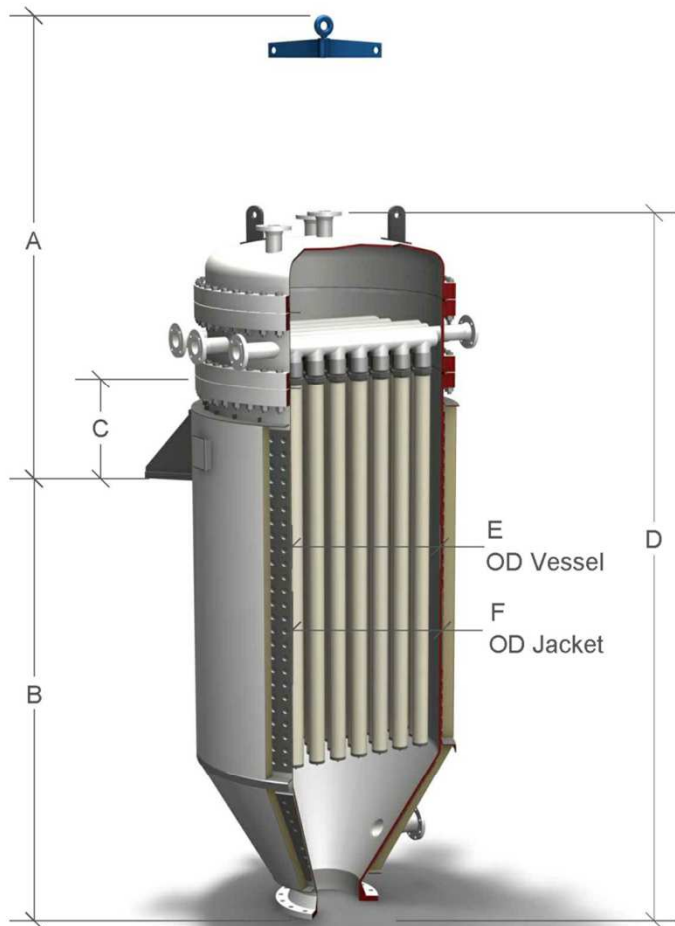
- In general, processes are divided into:
  - ▶ cold syngas cleaning
  - ▶ hot syngas cleaning
- Cold syngas cleaning – wet (absorption, scrubbing) or dry methods (adsorption....)
- Hot syngas cleaning – mostly dry methods (catalysis, adsorption...)
- A combination of hot and cold cleaning also possible
  - ▶ when switching to the cold method, the next hot step cannot be included

Why?

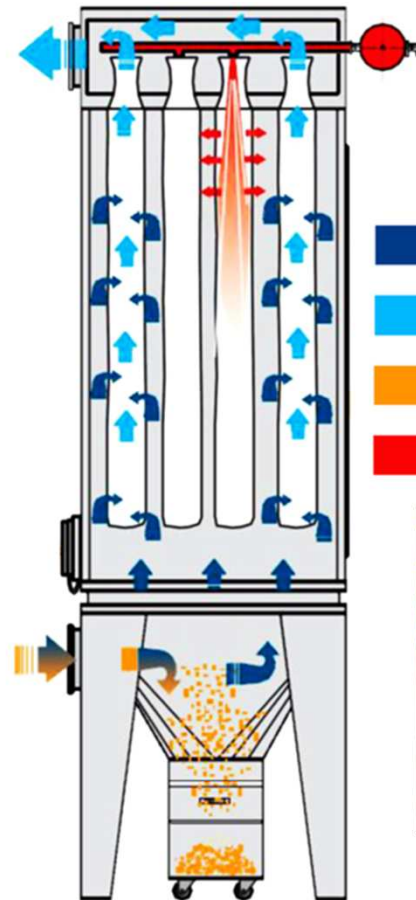
**cooling down** ↓ and **heating up** ↑ **again** makes no sense



- In general, PM removal processes are divided into:
  - ▶ dry methods
  - ▶ wet methods
- According to principle, PM removal processes are divided into:
  - ▶ inertial separation
    - cyclones
  - ▶ barrier filtration
    - ceramic candle filters
    - textile baghouse filters
    - granular bed filters
  - ▶ electrostatic separation
- +
- ▶ scrubbing (combined techniques for removal of PM + other specific contaminants)



Ceramic candle filter  
(up to 900 °C)



- Feed gas
- Clean gas
- Particulates
- Reverse pulse gas

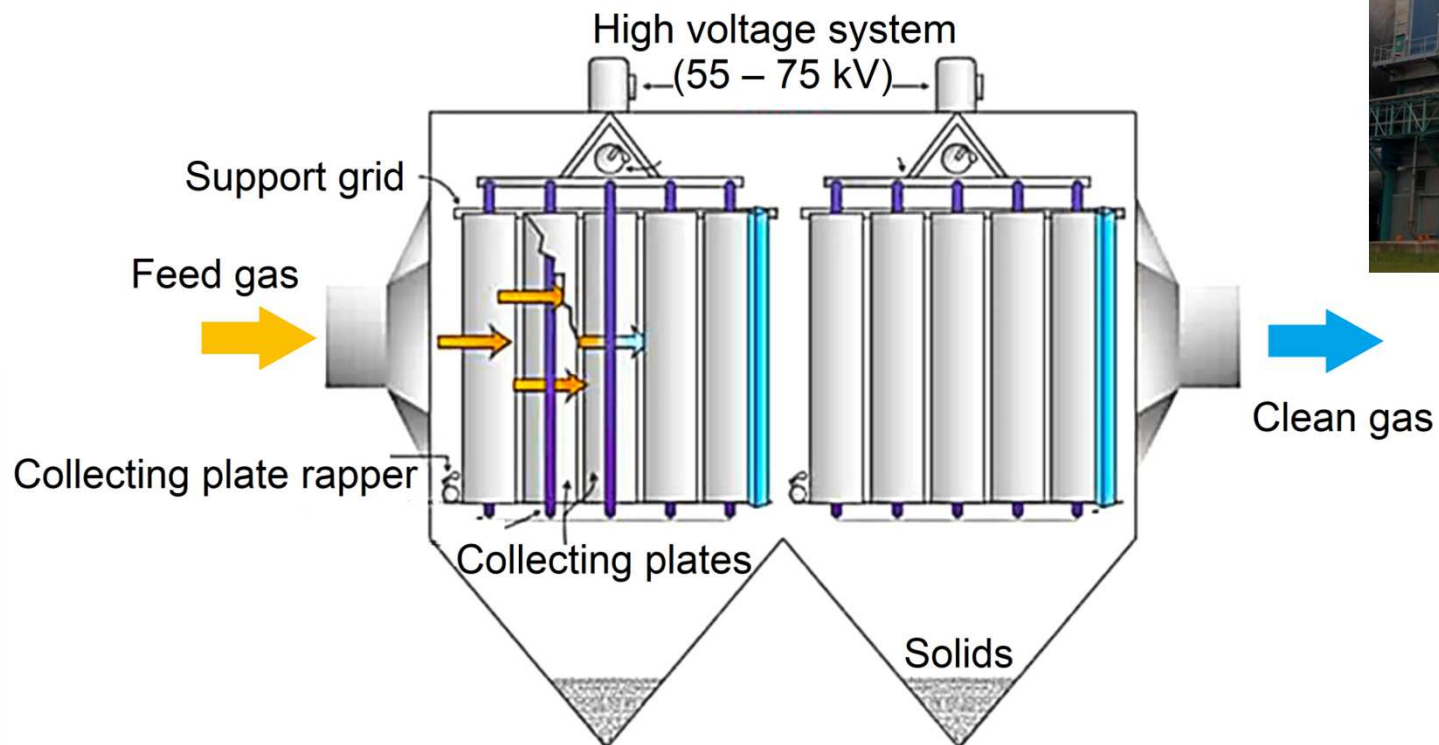


Teflon baghouse filter  
(up to 280 °C)



## ■ Electrostatic precipitator (ESP)

- ▶ PM transport between 2 electrodes
- ▶ internal (wire) electrode produces corona discharge
- ▶ no electric arc !





■ In general, tar removal processes are divided into:

▶ cold syngas cleaning = scrubbing with liquids

water

engine oil

diesel fuel

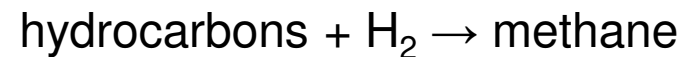
fatty acid methyl esters derived from vegetable oil

▶ hot syngas cleaning (metallic catalysts needed)

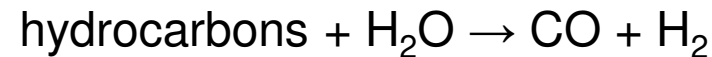
thermal cracking



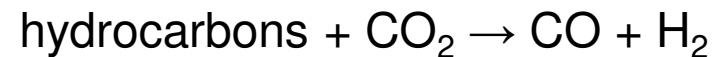
hydrocracking



steam reforming



dry reforming







■ In general, hydrogen halides removal processes are divided into:

▶ wet methods = scrubbing

water

NaOH solution

▶ dry methods = physical and chemical adsorption

sodium aluminate ( $\text{NaAlO}_4$ )

CaO

limestone ( $\text{CaCO}_3$ )

$\text{Na}_2\text{CO}_3$

$\text{NaHCO}_3$

$\text{K}_2\text{CO}_3$



- In general, denitrification processes are divided into:
  - ▶ cold processes = wet scrubbing
  - ▶ hot processes = catalytic thermal decomposition
  - ▶ hot or cold sorption (chemisorption as well as physical adsorption)



## ■ Hot processes

- ▶ adsorption primarily chemical reaction with metal oxides  
 $\text{Fe}_2\text{O}_3$ ,  $\text{ZnO}$ ,  $\text{Mn}_2\text{O}_3$ ,  $\text{CuO}$ ,  $\text{CaO}$
- ▶ hot catalysis e.g.  $10\text{Ni}-1\text{Nb}-1\text{Ca}/\text{Al}_2\text{O}_3$   
sometimes not clear if the process is adsorptive or catalytic
- ▶ in situ addition of alkaline compounds  
limestone for fluidized bed;  $\text{K}_2\text{CO}_3$  for hydrothermal gasification
- ▶ feeding material impregnation (coal, biomass, waste ...)  
wet impregnation with  $\text{Na}_2\text{CO}_3$  before gasification

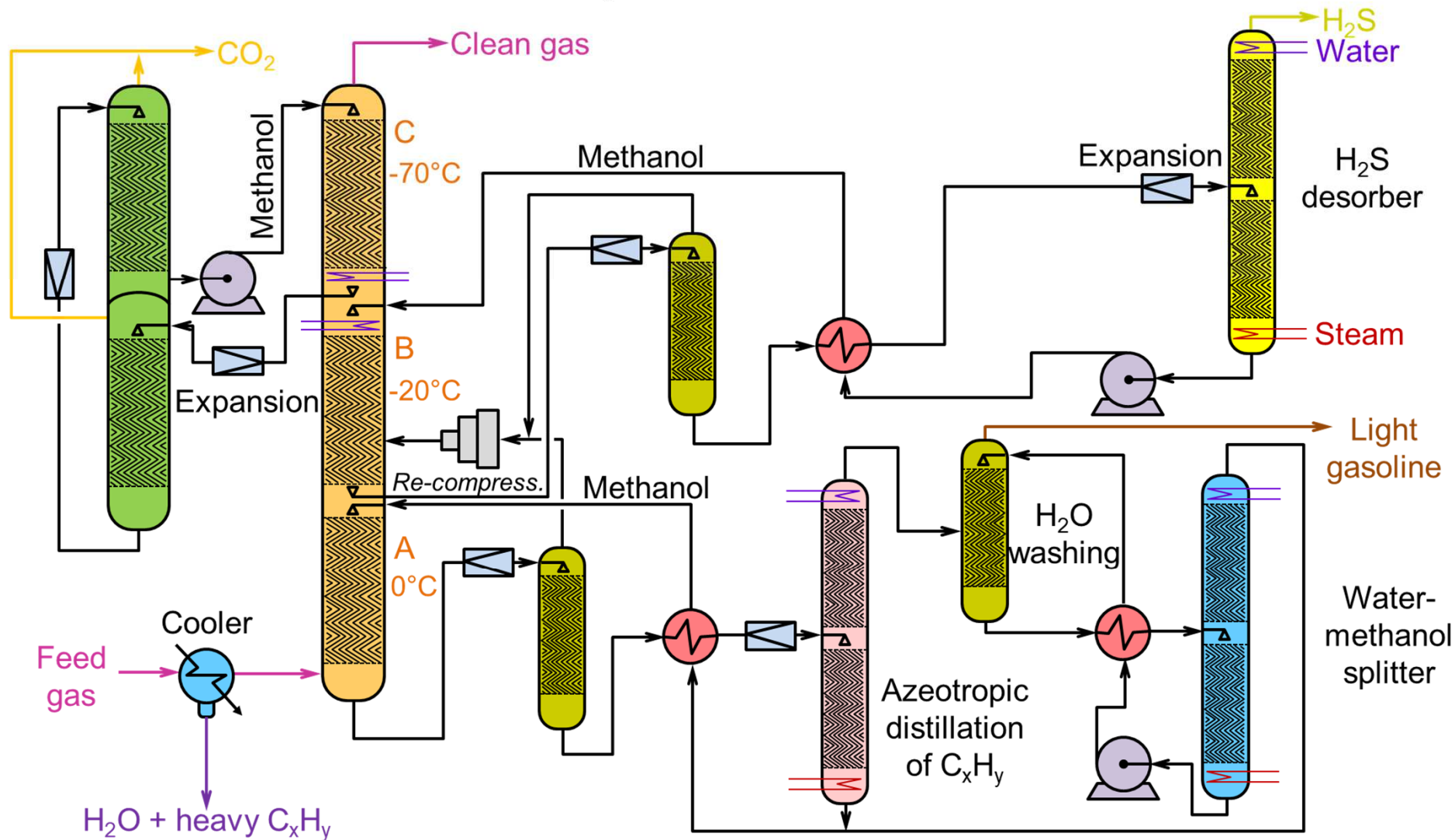
## ■ Cold processes

- ▶ cryogenic distillation
- ▶ cold catalysis system Lo-Cat™ with chelated Fe
- ▶ oxidative desulfurization  
conversion of  $\text{H}_2\text{S}$  to elemental S using polyoxometalate
- ▶ absorption (see the next slide)

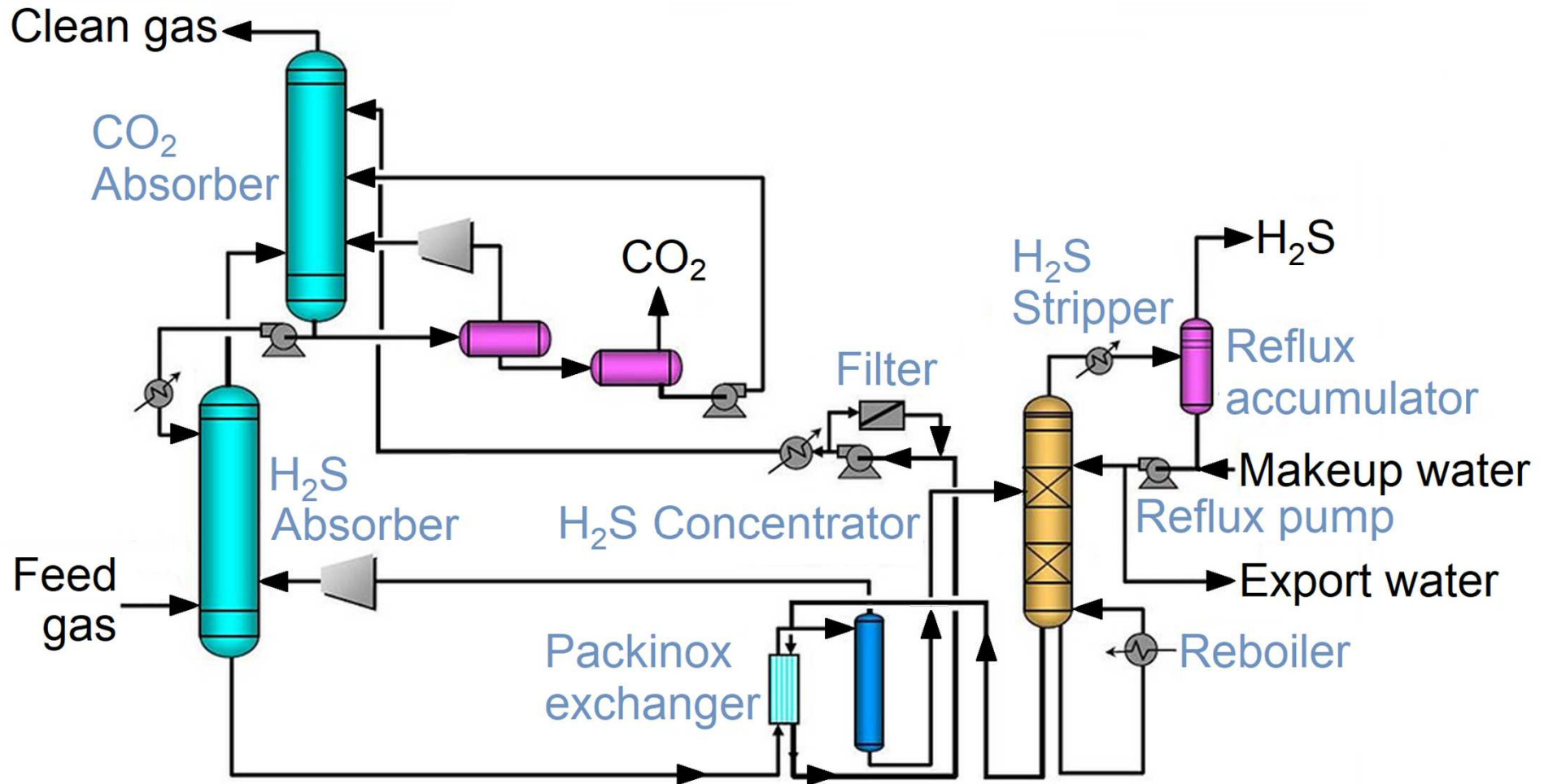
Wet processes overview (specific design enables also CO<sub>2</sub> separation)

Absorption type	Solvent	Commercial name
Physical absorption	Polyethylene glycol alkyl ethers	Selexol, Coastal AGR, Genosorb
	N-methyl-2-pyrrolidone	Purisol
	Methanol	Rectisol, Ifpexol
	Propylene carbonate	Jeffsol
Chemical absorption	Primary amines (MEA, DGA)	
	Secondary amines (DEA, DIPA)	
	Tertiary amines (MDEA, TEA)	
	Potassium carbonate	
Hybrid absorption	MEA + sulfolane	Sulfinol-M
	DIPA + sulfolane	Sulfinol-D
	Methanol + MDEA	Amisol
	Sterically hindered amines	Flexsorb

High pressure method: **A** –  $C_xH_y$  removal, **B** –  $H_2S$  removal, **C** –  $CO_2$  removal



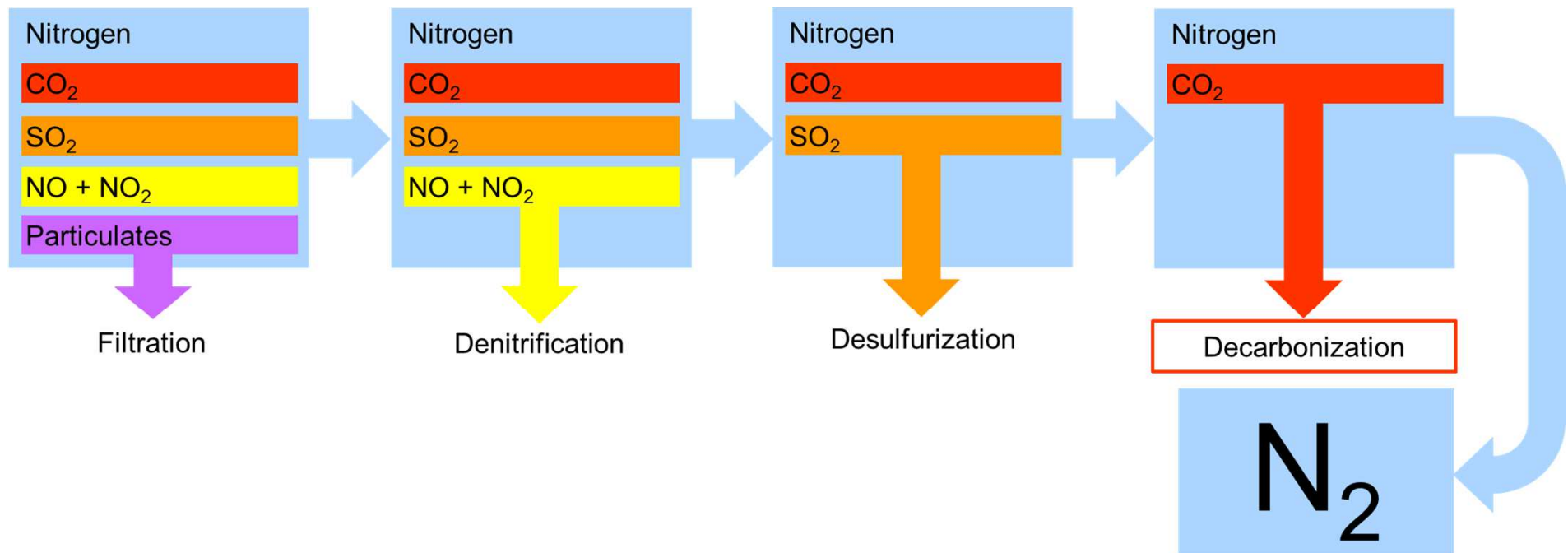
High pressure method (2 – 14 MPa) for acid gases removal  $\text{H}_2\text{S}$  and  $\text{CO}_2$



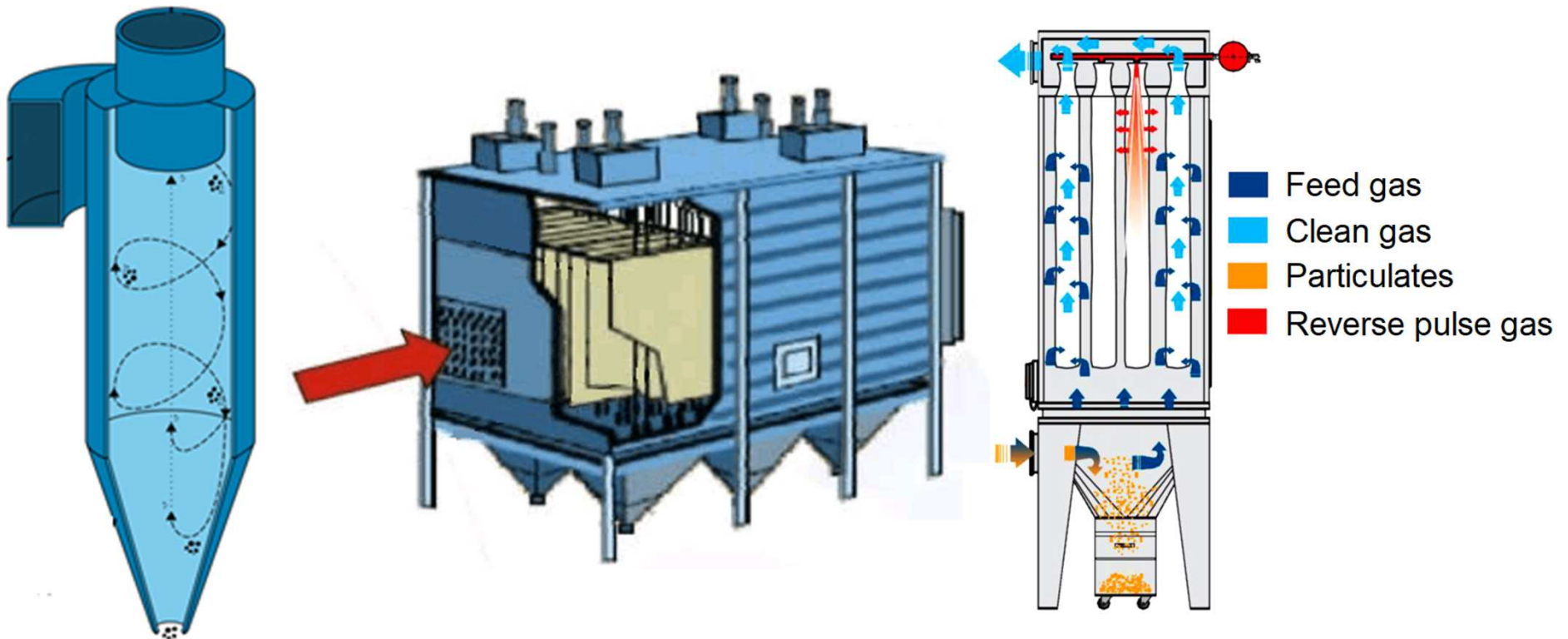




- A typical sequence of flue gas cleaning steps
  - ▶ lower number of contaminants compared to pre-combustion processes
  - ▶ but huge volume flows

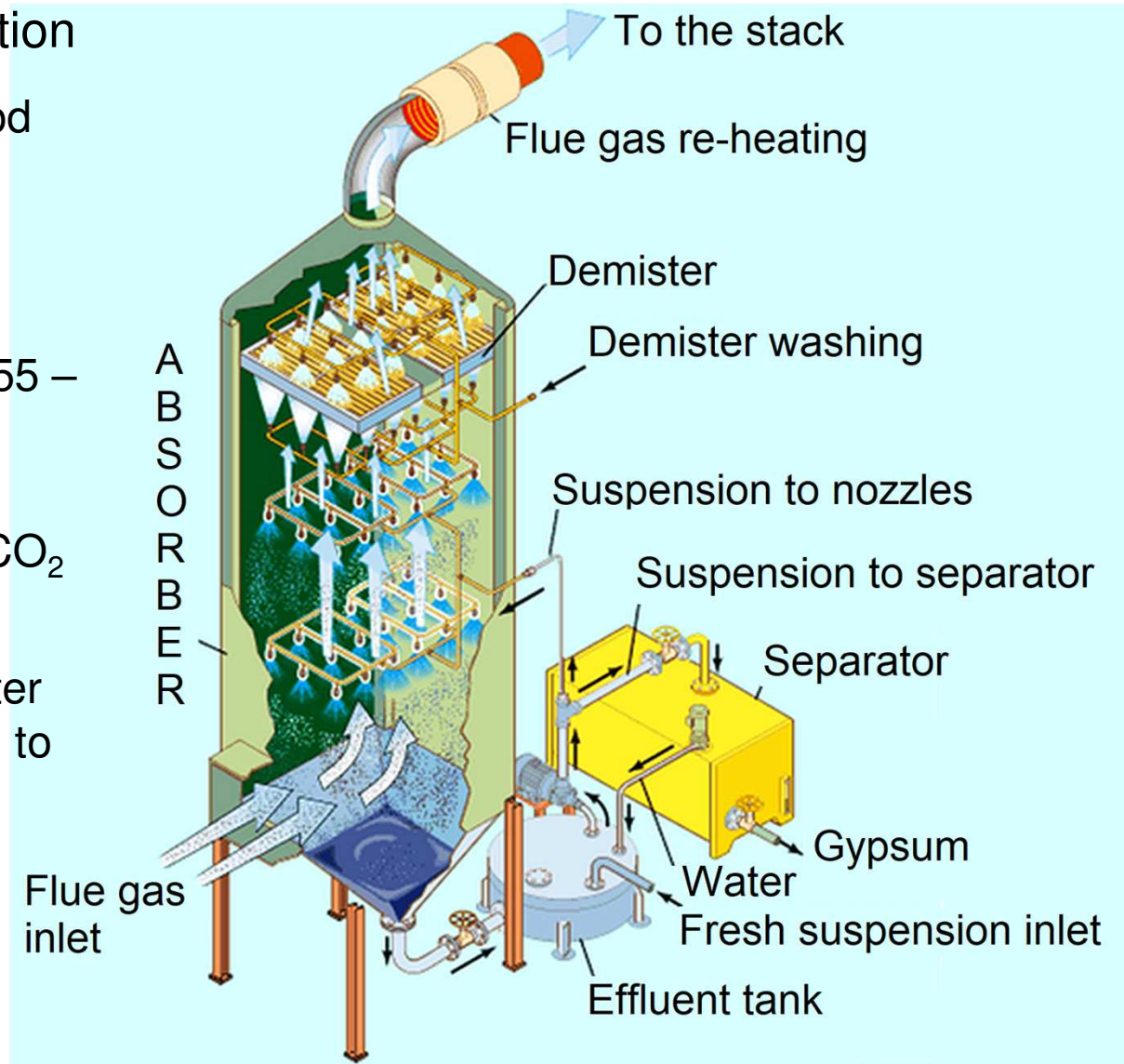


- Basically, only 3 methods widespread in the industrial practice
  - ▶ cyclones (typical pre-separators for fluidized bed combustors)
  - ▶ electrostatic separation (precipitation), ESP
  - ▶ barrier filtration – baghouse filters



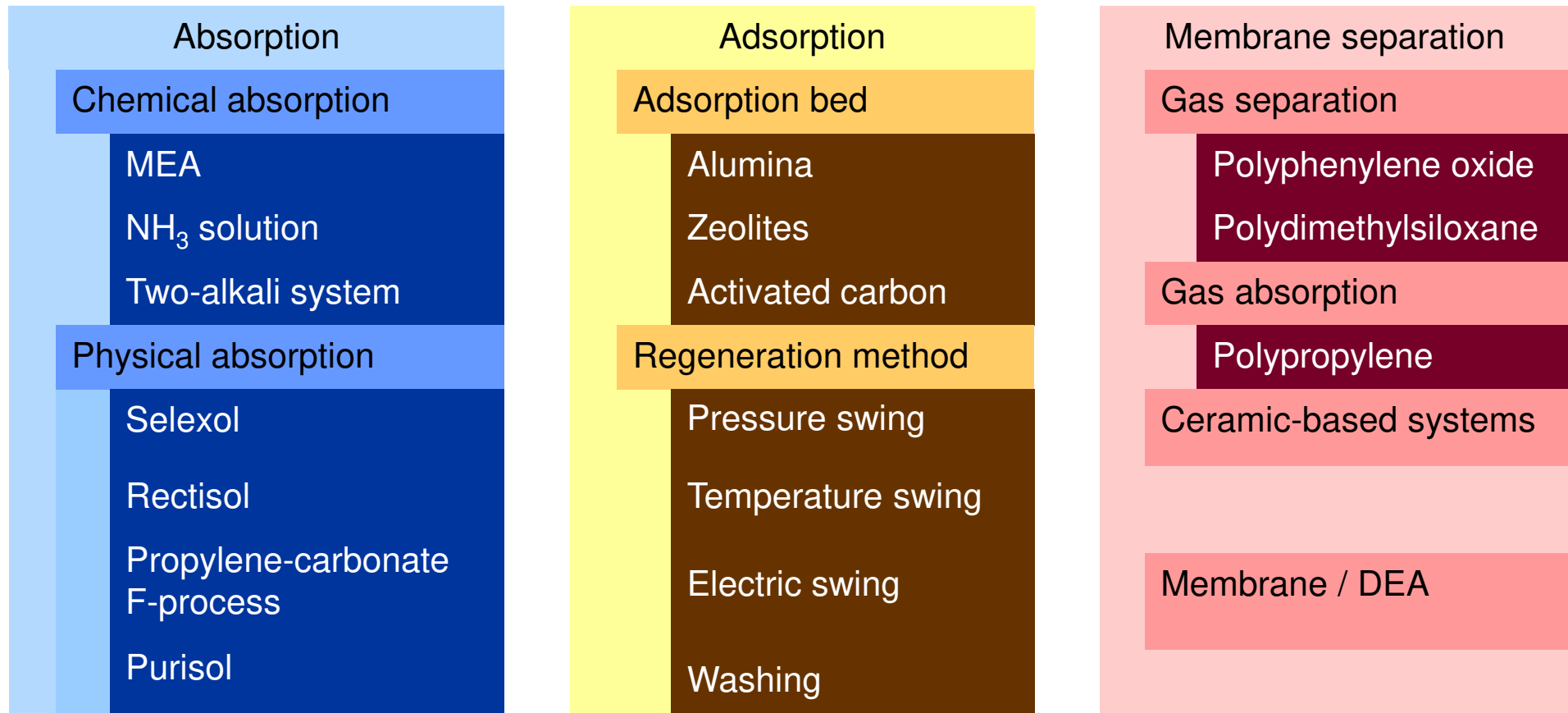
## Wet flue gas desulfurization

- ▶ the most frequent method
- ▶ except fluidized bed combustors
- ▶ inlet  $t = 110 - 140\text{ }^{\circ}\text{C}$
- ▶ optimal  $t$  for absorption  $55 - 60\text{ }^{\circ}\text{C}$
- ▶ Based on reaction
$$\text{SO}_2 + \text{CaCO}_3 \rightarrow \text{CaSO}_3 + \text{CO}_2$$
$$2\text{CaSO}_3 + \text{O}_2 \rightarrow 2\text{CaSO}_4$$
- ▶ After passing the demister clean flue gas re-heated to ca.  $75 - 95\text{ }^{\circ}\text{C}$

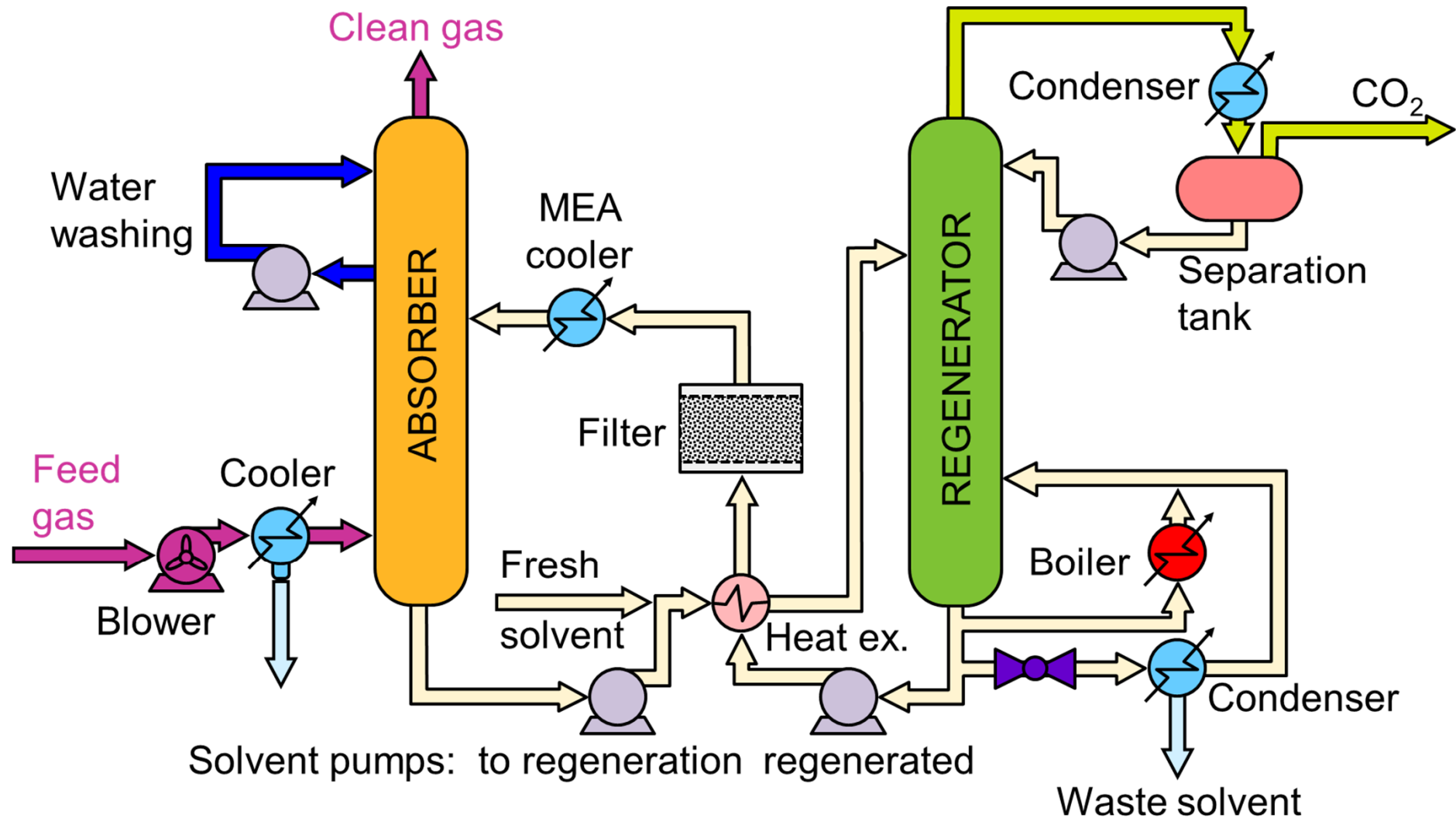


- Note: Some processes also common to desulfurization

### 3 main ways



Very similar arrangement for wide range of amine-based solvents





- Absorption columns filled with various packing materials (to enhance the mass exchanging surface)



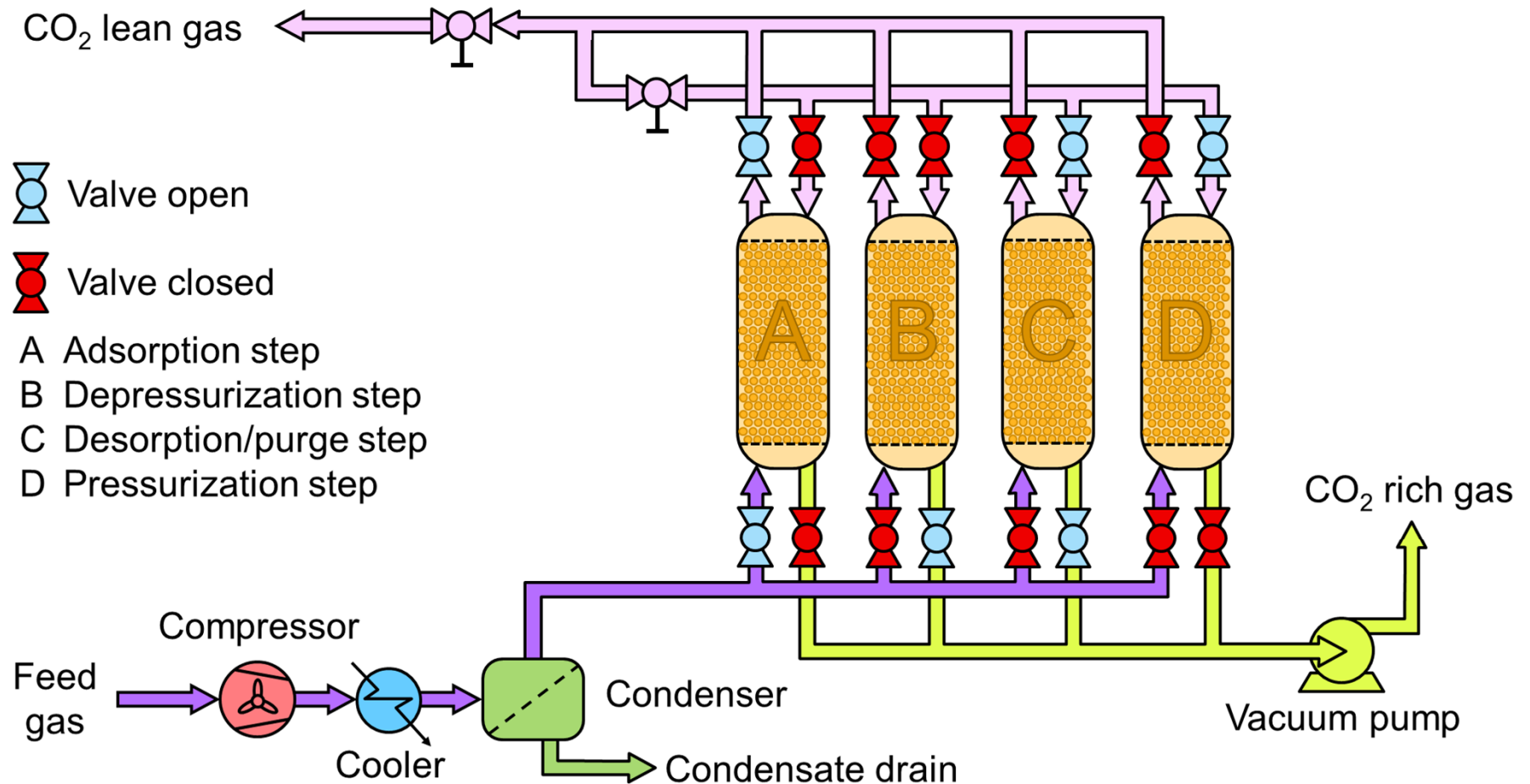
Structured packing



Various types of bulk materials (Raschig rings etc.)



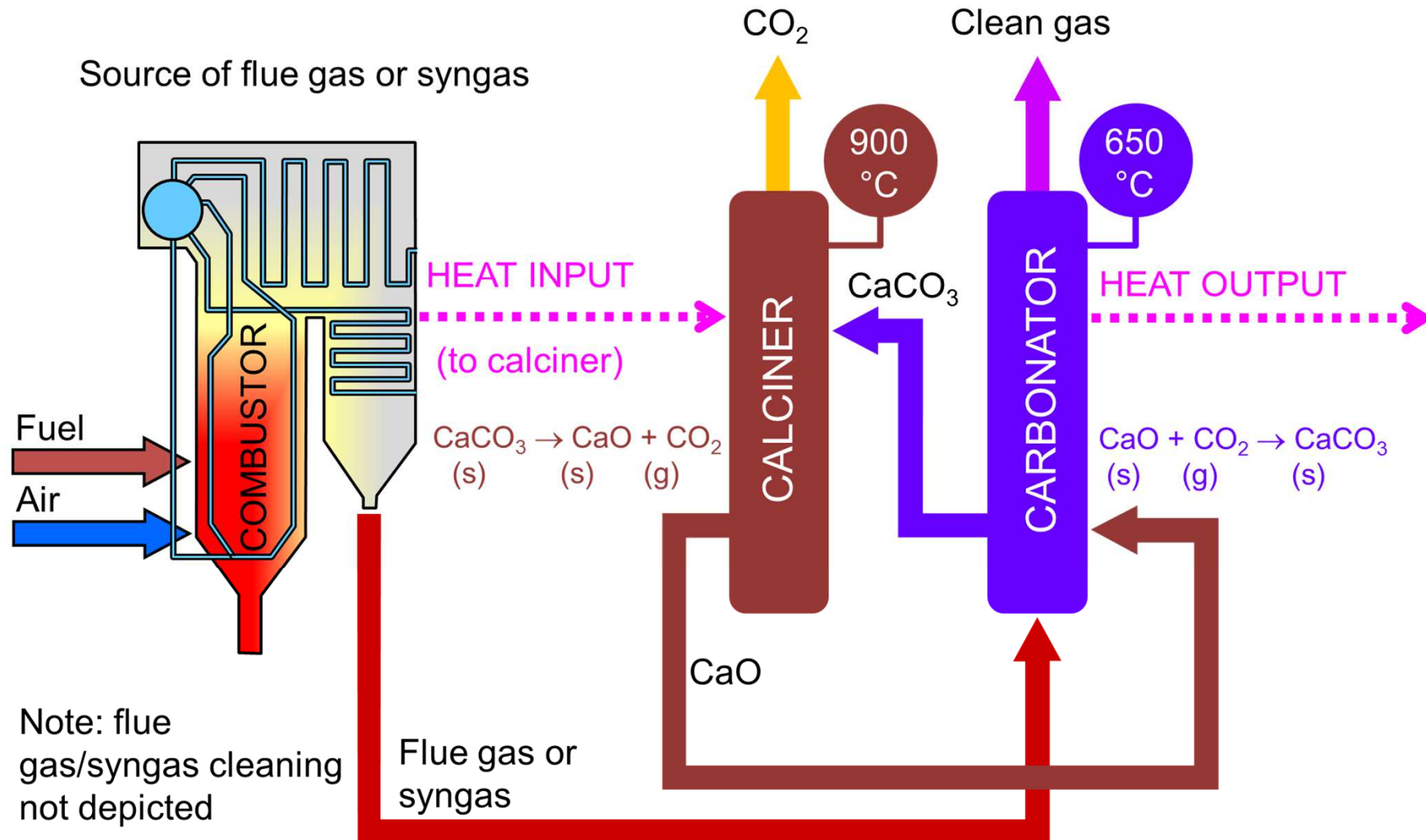
General scheme common for many gas mixtures – specific conditions and sorbents required



- Industrial scale unit Pioneer Technology Co.
  - ▶ Applied zeolites 5A, 13X, APGIII and activated carbon
  - ▶ H<sub>2</sub>O and H<sub>2</sub>S removed before PSA, CO<sub>2</sub> content in feed gas 13 – 15%
  - ▶ CO<sub>2</sub> removal efficiency 74 – 98%, CO<sub>2</sub> output purity 94 – 96%
  - ▶ adsorption pressure 1 – 3.5 bar / regeneration pressure 0.02 – 0.15 bar



Special example of the high-temperature chemical adsorption process

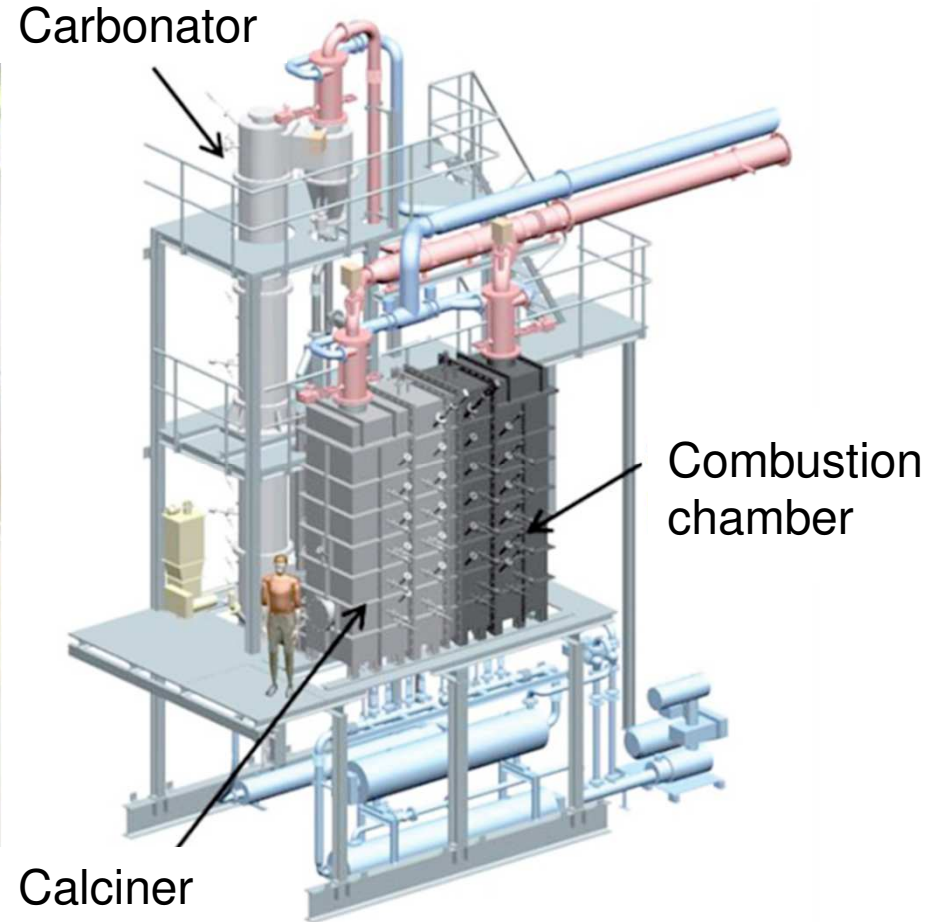




- Pilot scale unit: Technische Universität Darmstadt
  - ▶ circulating adsorbent:  $\text{CaCO}_3$  rich limestone
  - ▶ adsorption (carbonation)  $t = 650\text{ }^\circ\text{C}$ ; regeneration (calcination)  $t = 950\text{ }^\circ\text{C}$



Carbonator



Combustion chamber

Calciner



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5. Riboldi L., Bolland O. Energy Procedia 114 (2017) 2390 – 2400
6. Staf M., Ciahotný K., Krtková E. Regenerativní záchyt oxidu uhličitého z energetických procesů. Zpravodaj hnědé uhlí (2017)
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