

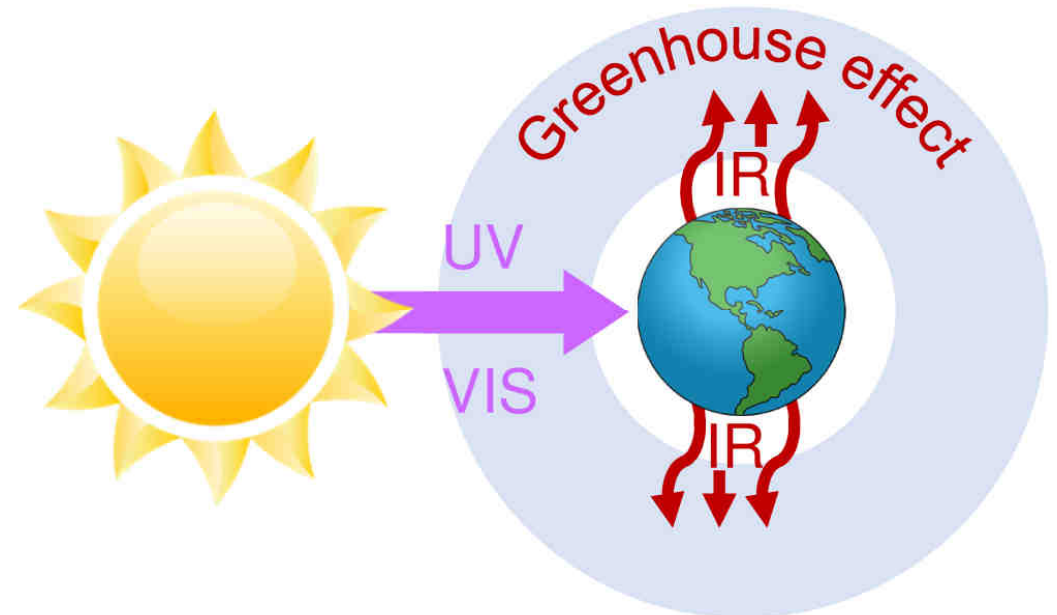
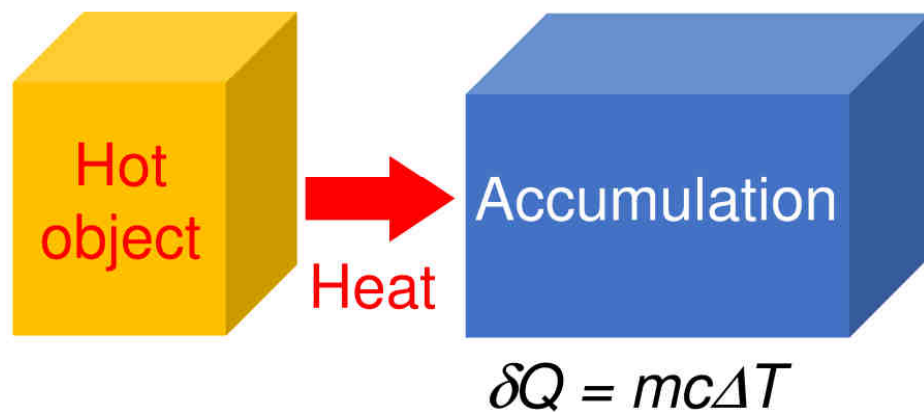


# Greenhouse Gases Mitigation CO<sub>2</sub> Capture and Utilization

Topic No: 1

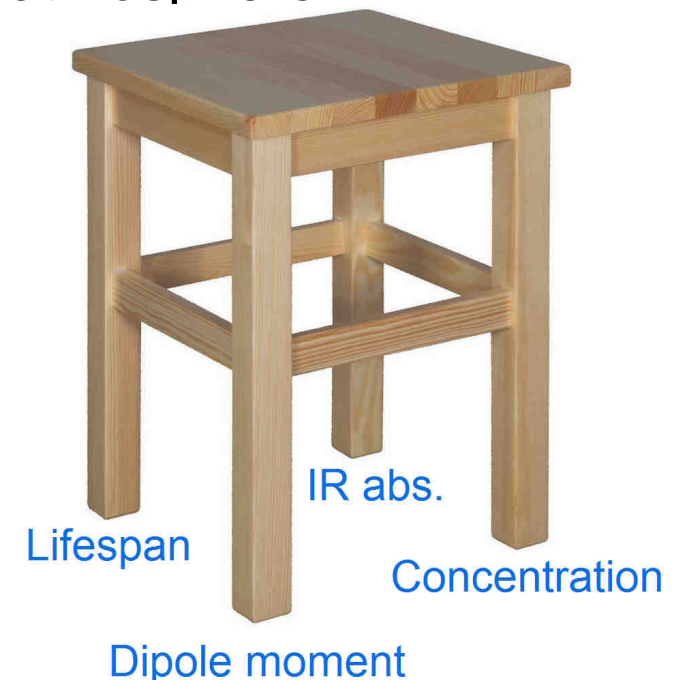
1. Mechanism of the greenhouse effect
2. Overview of greenhouse gases and their groups
3. Physico-chemical properties of greenhouse gases
4. Energy balance in the atmosphere, radiative forcing and global warming potential
5. Climate theory
6. Overview of economic sectors contributing to GHG emissions

- The greenhouse effect **is not** simply the accumulation of heat.
- Principle: The Earth absorbs UV and VIS and emits IR, but specific gases retain it in the atmosphere.
- Gases capable of this process = greenhouse gases (GHGs).





- GHG must absorb radiation in IR part of spectrum.
- Molecules of GHG must change their dipole moment due to IR absorption
  - ▶ Symmetric di-atomic molecules ( $H_2$ ,  $N_2$ ,  $O_2$ ) do not change their dipole moment  $\Rightarrow$  IR inactive;
  - ▶ Molecules with different partial charges on the atoms ( $CO$ ,  $CO_2$ ,  $N_2O$ ...) change the dipole moment  $\Rightarrow$  IR active.
- GHG molecules must have sufficient lifetime in the atmosphere.
  - ▶ e.g.  $CO_2$  50 – 200 years,  $CH_4$  12 – 17 years etc.
- GHG must be present at sufficient concentrations
  - ▶ e.g.  $CO_2 > 400$  ppmv,  $H_2O \cong 0.4$  % vol.



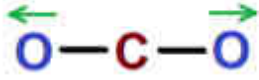
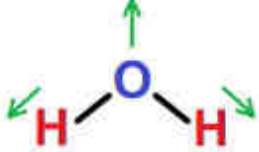
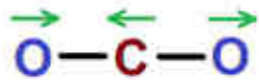
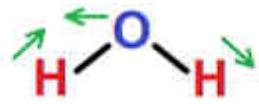
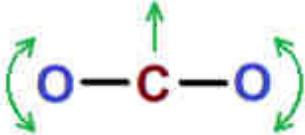
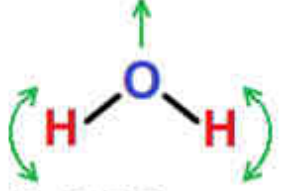
Hundreds of compounds ... the main are:

- H<sub>2</sub>O (vapor)
- CO<sub>2</sub>
- C<sub>x</sub>H<sub>y</sub> (especially CH<sub>4</sub>)
- N<sub>2</sub>O
- F-gases a ClF-gases:
  - ▶ CFCs (chlorofluorocarbons)
  - ▶ HFCs (hydrofluorocarbons)
  - ▶ PFCs (perfluorocarbons)
  - ▶ SF<sub>6</sub> (sulfur fluoride)
- O<sub>3</sub>

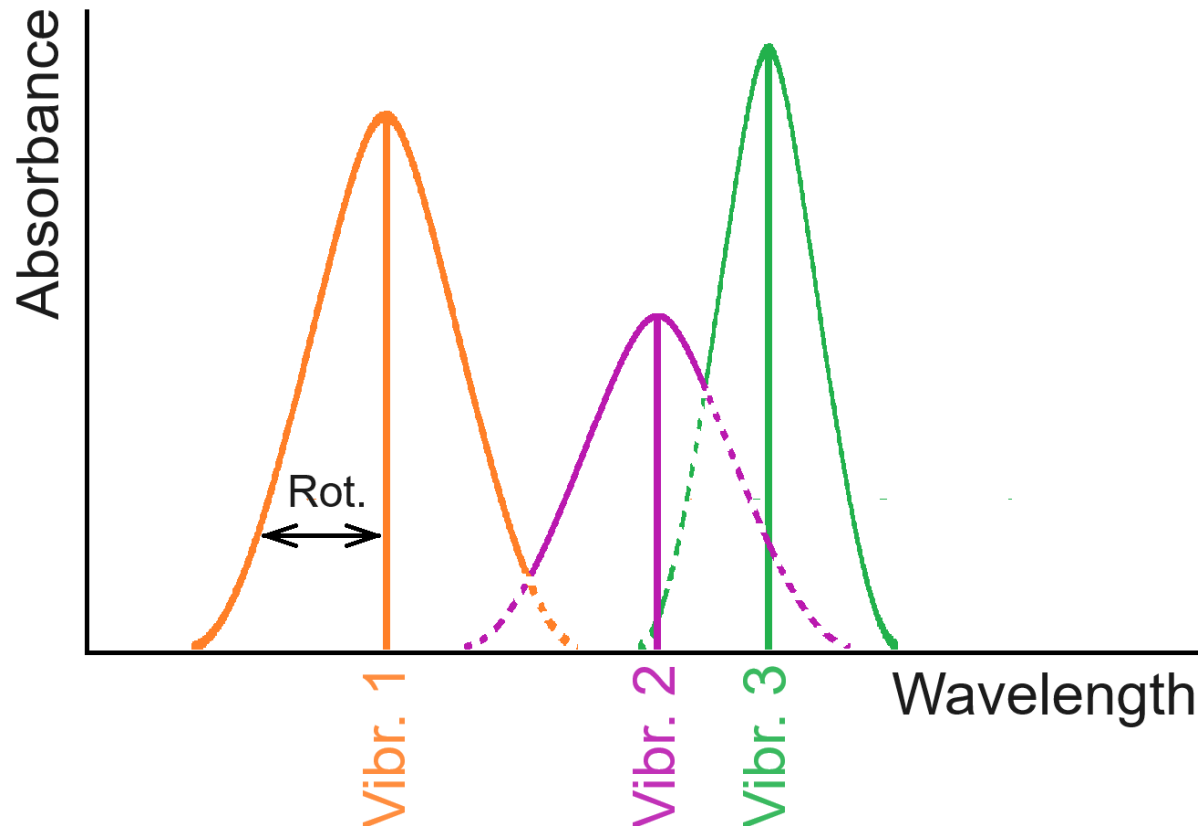
What is their origin:

■ H <sub>2</sub> O (vapor)	NATURAL	
■ CO <sub>2</sub>	NATURAL	ANTHROPOGENIC
■ C <sub>x</sub> H <sub>y</sub> (especially CH <sub>4</sub> )	NATURAL	ANTHROPOGENIC
■ N <sub>2</sub> O	NATURAL	ANTHROPOGENIC
■ F-gases a ClF-gases:		
▶ CFCs (chlorofluorocarbons)		ANTHROPOGENIC
▶ HFCs (hydrofluorocarbons)		ANTHROPOGENIC
▶ PFCs (perfluorocarbons)		ANTHROPOGENIC
▶ SF <sub>6</sub> (sulfur fluoride)		ANTHROPOGENIC
■ O <sub>3</sub>	NATURAL	ANTHROPOGENIC

- Absorption of light induces molecular vibrations.
- Quantum transition during IR absorption = values of molecular vibrations
  - ▶ Each molecular vibration has its specific wavelength value

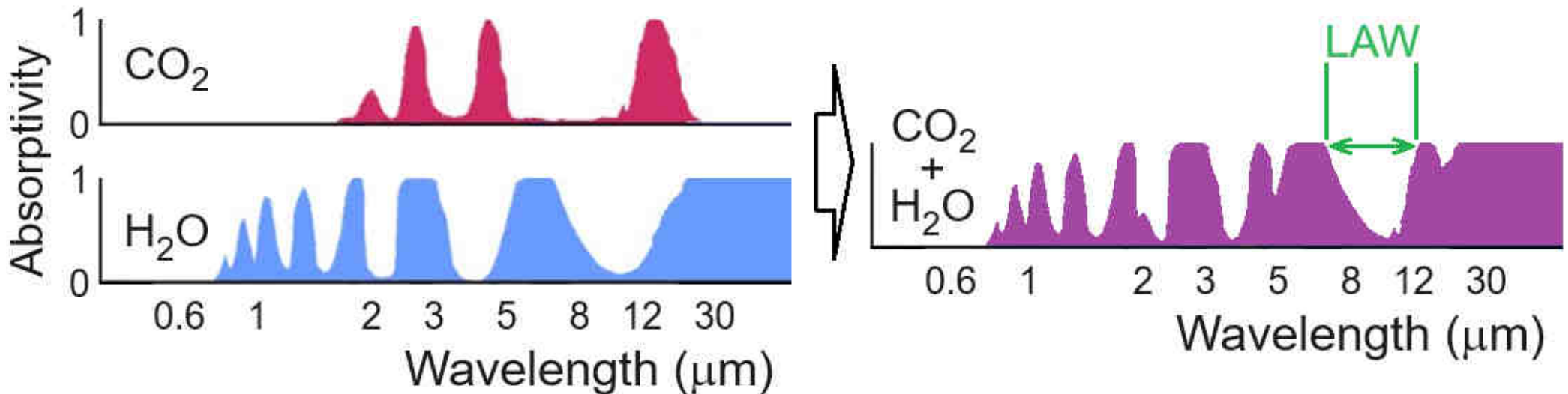
	Carbon dioxide	Water
Symmetric stretch ( $\nu_s$ )	 7,490 nm	 2,738 nm
Asymmetric stretch ( $\nu_{as}$ )	 4,257 nm	 2,656 nm
Bend ( $\delta$ )	 14,992 nm	 6,269 nm

- Absorption of light induces molecular vibrations.
- Quantum transition during IR absorption = values of molecular vibrations
  - ▶ Each molecular vibration has its specific wavelength value
  - ▶ But 1 molecular vibration induces high number of various rotation levels ⇒ extension of absorption belt width ⇒ continuous spectrum



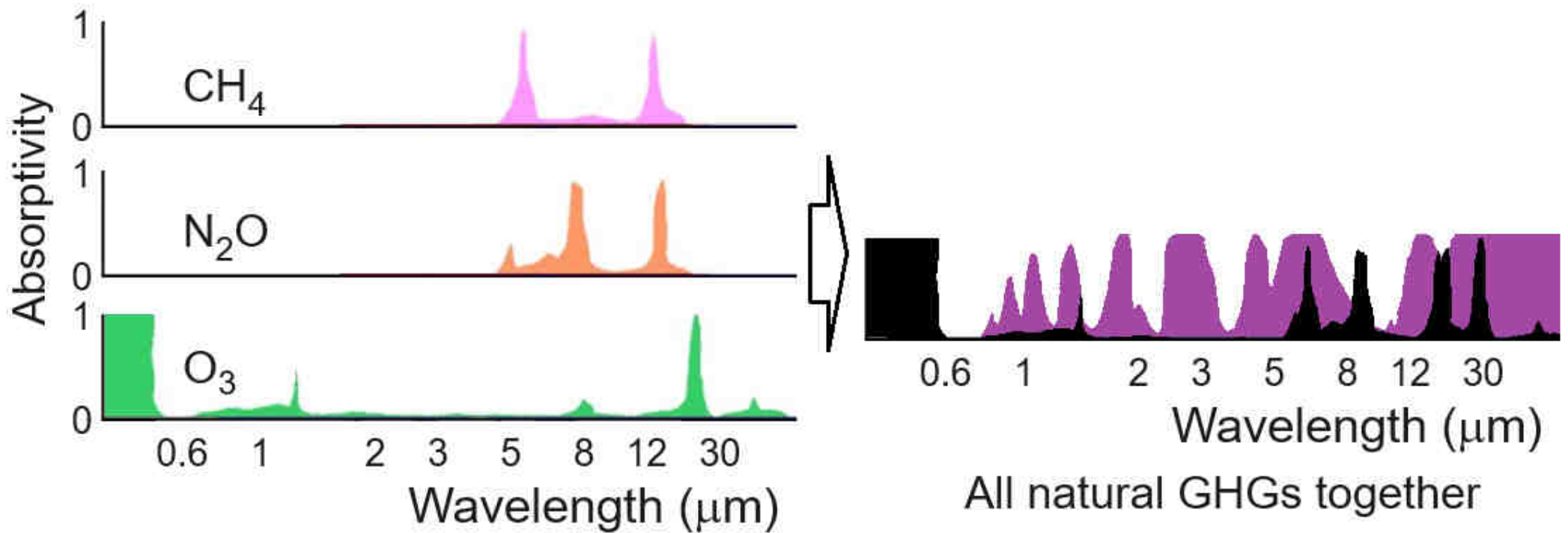


- CO<sub>2</sub> and H<sub>2</sub>O – the most concentrated GHGs in the atmosphere
  - ▶ e.g. CO<sub>2</sub> concentration more than 200 times higher than CH<sub>4</sub>
- Cooling of the atmosphere strongly limited by the CO<sub>2</sub> + H<sub>2</sub>O absorption bands.
- Where neither CO<sub>2</sub> nor H<sub>2</sub>O absorbs, the Earth well radiates IR into space.

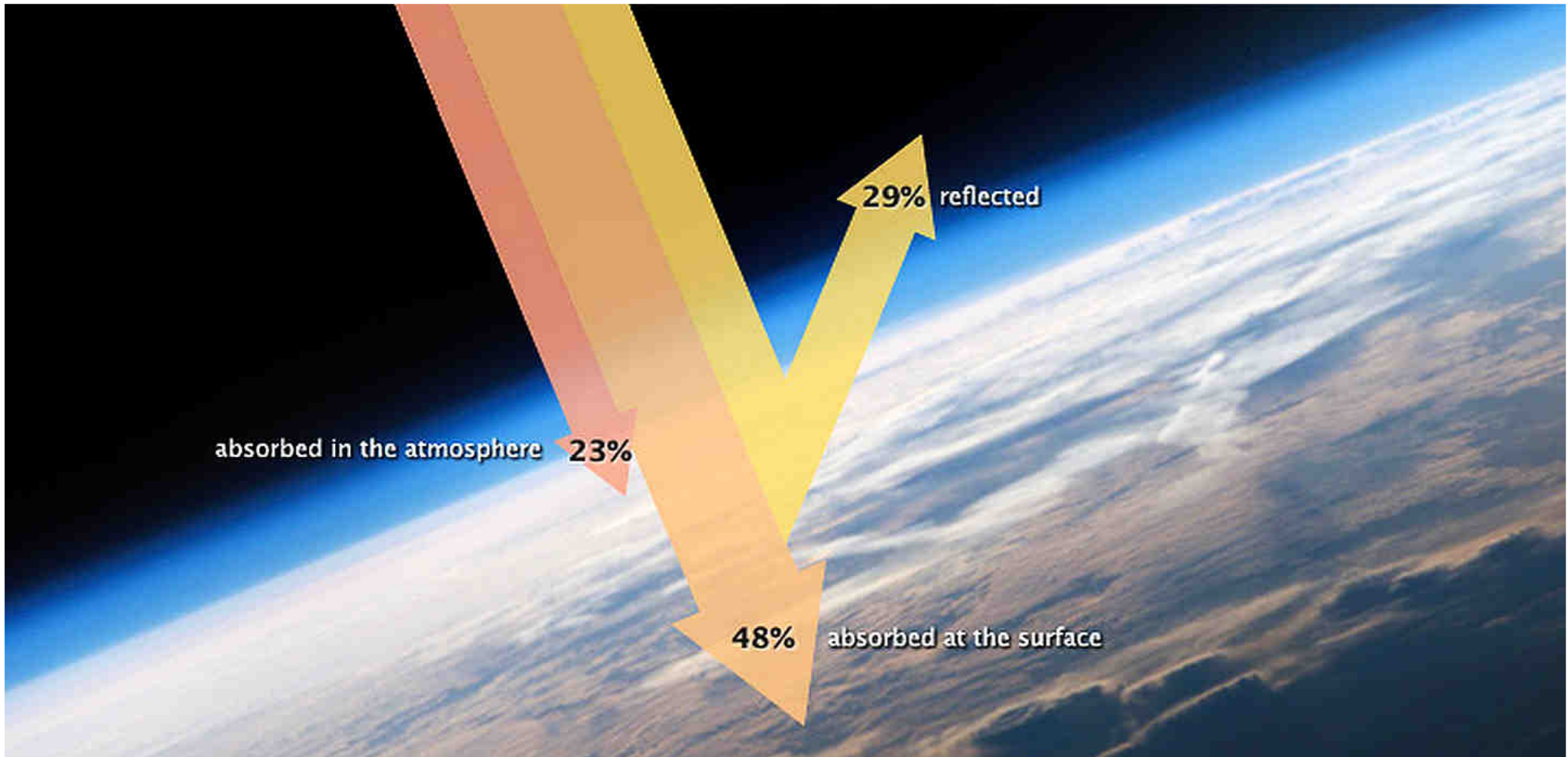


- YES: among natural GHGs, e.g.  $N_2O$  shows maximum within the LAW.
- Especially solely anthropogenic  $CFC_s$ ,  $HFC_s$ ,  $PFC_s$  absorb inside the LAW.

A compound absorbing within the LAW is more dangerous than  $CO_2$  and  $H_2O$ .

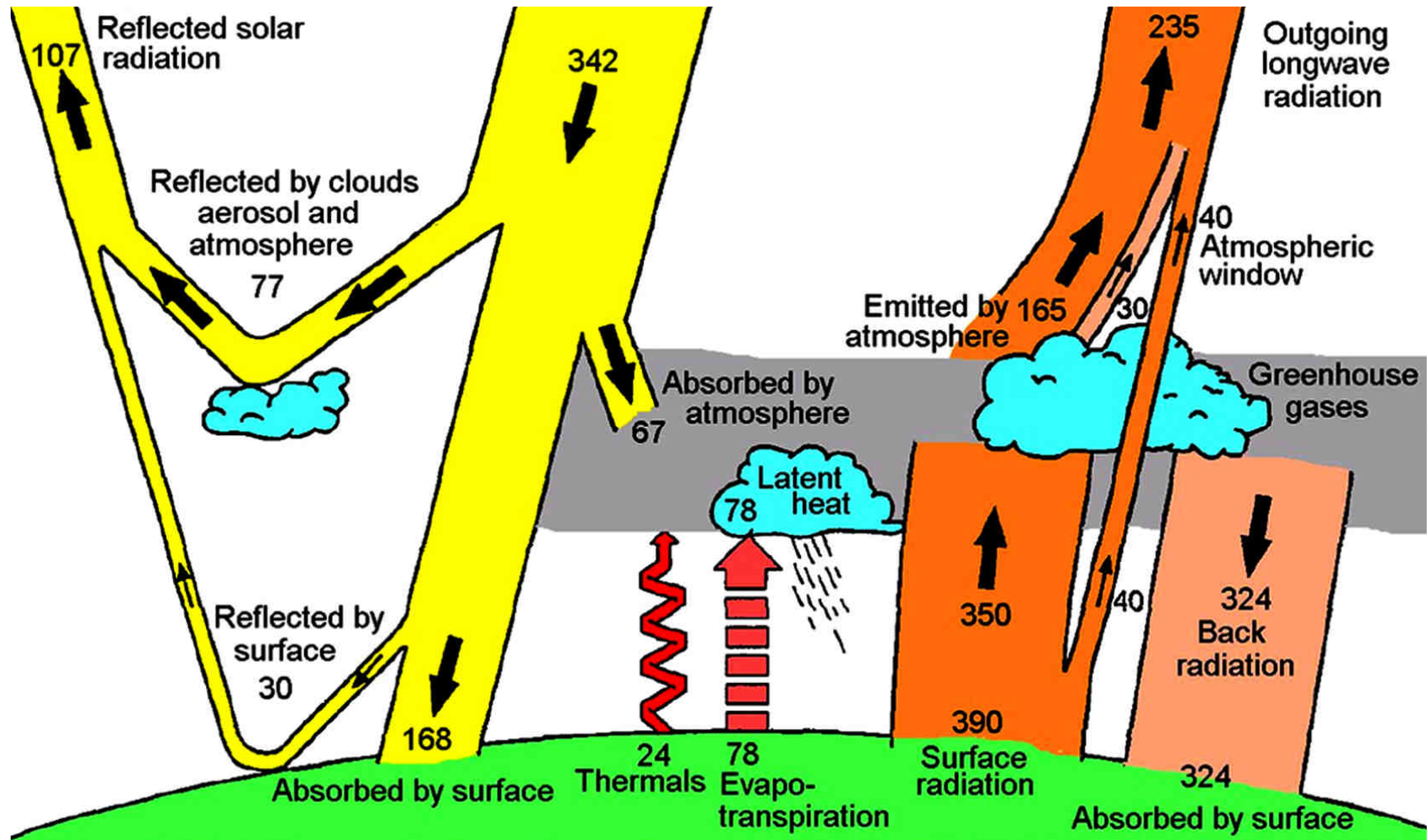


- Balance of incoming solar radiation (UV + VIS)
  - ▶ Total incoming energy flux  $\cong 340 \text{ W m}^{-2}$  (defined in tropopause)
  - ▶ 29% reflected ( $98.6 \text{ W m}^{-2}$ ), 23% ( $78.2 \text{ W m}^{-2}$ ) absorbed in the atmosphere



## ■ The overall energy balance of the Earth

- ▶ The atmospheric window shown at 38% proportion of clear sky
- ▶  $30 \text{ W m}^{-2}$  = radiative energy of clouds in longwave area





- There is the equilibrium between UV and visible radiation absorbed by the planet and reflection of IR radiation back to the space.
- Due to absorption of IR radiation, GHG gases change this ratio ⇒ accumulation of energy.
  - ▶ Since the beginning of the industrial era: Of all the investigated factors, only the concentration of GHGs changed along with the rise in temperature.

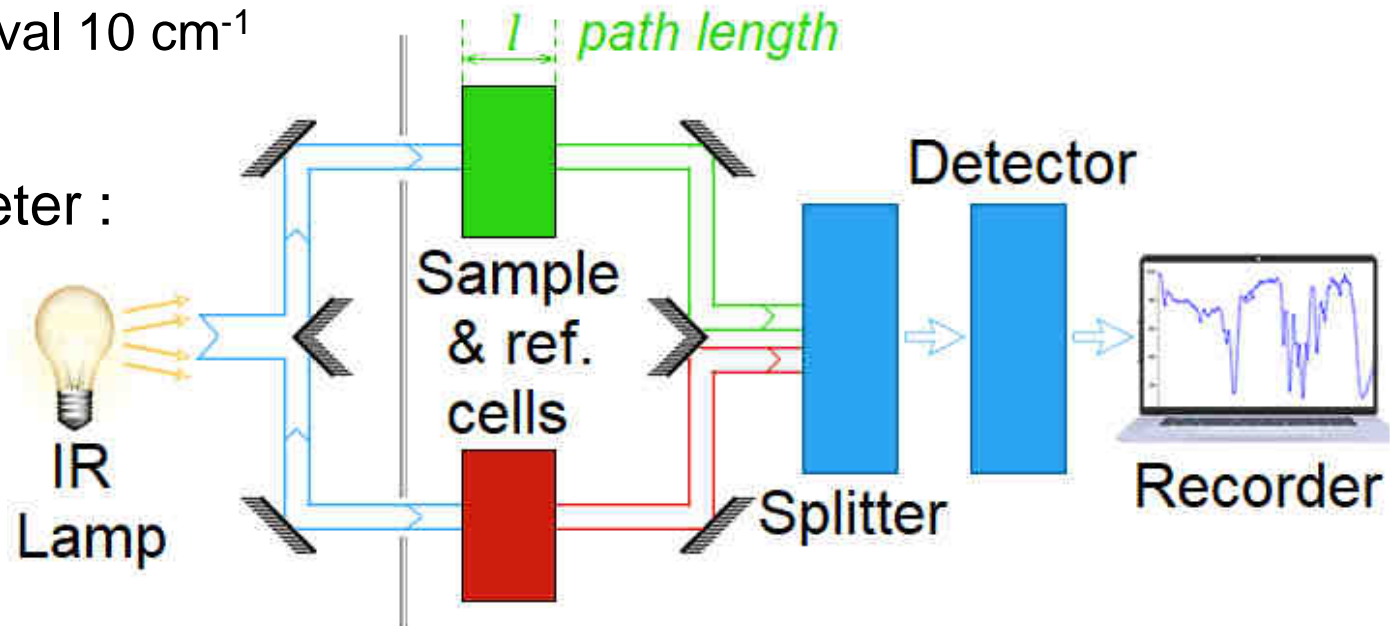
- Several possible parameters, which the most widespread are:
  - ▶ Radiative Forcing Capacity (RFC)
  - ▶ Global Warming Potential (GWP)
- RFC = the amount of energy per unit area per unit time, absorbed by greenhouse gases, which would otherwise be radiated into space
  - ▶ Do not confuse with „Radiative Forcing“ = difference between the solar energy absorbed by the Earth and the energy radiated back to outer space.
- GWP is a relative measure of how much heat is retained in the atmosphere by a gas;
  - ▶ GWP compares the amount of heat, retained by the certain amount of the particular gas, relative to the same amount of the reference gas - CO<sub>2</sub>
  - ▶ GWP is a dimensionless factor
  - ▶ GWP is related to CO<sub>2</sub>, thus  $GWP(\text{CO}_2) = 1$

- RFC expressed by the formula (Beer's law) :

$$RF = \sum_{n=1}^{100} \frac{Abs_i \cdot F_i}{l \cdot n}$$

- ▶  $Abs_i$  = integrated infrared absorbance in  $i^{th}$  interval
- ▶  $F_i$  = radiative forcing in  $i^{th}$  interval
- ▶  $l$  = path length of the IR measuring cell (cm)
- ▶  $n$  = number density of GHG molecules ( $cm^{-3}$ )
- ▶ Subscript  $i$  = interval  $10\ cm^{-1}$

- RFC determined using the IR spectrometer :



- GWP values published by the Intergovernmental Panel on Climate Change (IPCC)
- GWP changed several times between 1996 and 2001.
- In 2001, the exact formula published in the third IPCC report:

$$GWP(x) = \frac{\int_0^{TH} a_x \cdot [x(t)] dt}{\int_0^{TH} a_r \cdot [r(t)] dt}$$

- ▶  $TH$  = time horizon, for the calculation (20, 100 or 500 years)
- ▶  $a_x$  = radiative efficiency for unit increase of atmospheric abundance of the selected substance ( $W m^{-2} kg^{-1}$ )
- ▶  $[x(t)]$  = time-dependent decay of the substance (decrease of its abundance from its release in the time  $t = 0$  until  $t = TH$ )
- ▶ Denominator of the fraction includes the same variables for the reference gas.



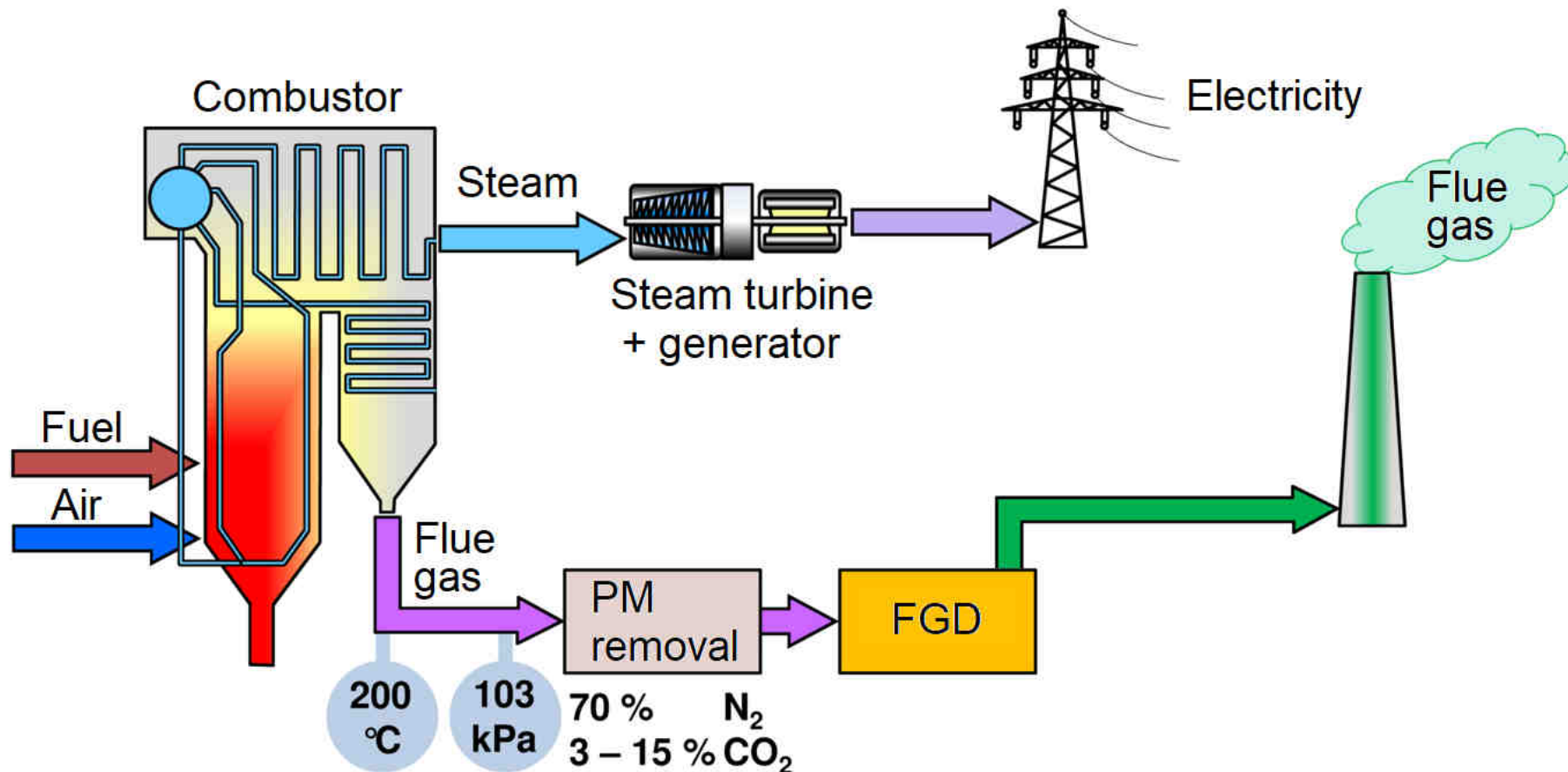
- GWP depends on the following factors:
  - ▶ The rate of absorption of IR radiation by the substance;
  - ▶ Position of wavelengths, absorbed by the substance, in the solar spectrum;
  - ▶ Lifetime of the substance in the atmosphere.
- GWP calculation meets problems:
  - ▶ Radiative efficiencies  $a_x$ ,  $a_r$  not constant within the whole-time horizon
  - ▶ For the majority of gases IR absorbance increases linearly with their abundance in the atmosphere
  - ▶ Several important GHGs show non-linear dependence, e.g. CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O
  - ▶ Increase of CO<sub>2</sub> concentrations has lower impact on overall IR absorption (saturation of corresponding wavelengths) TOO HIGH CONCENTRATION
  - ▶ Calculation for H<sub>2</sub>O almost impossible: Unequal H<sub>2</sub>O distribution in troposphere (average ca. 0.4 % vol., but up to 1.8 % vol. near the sea level.)



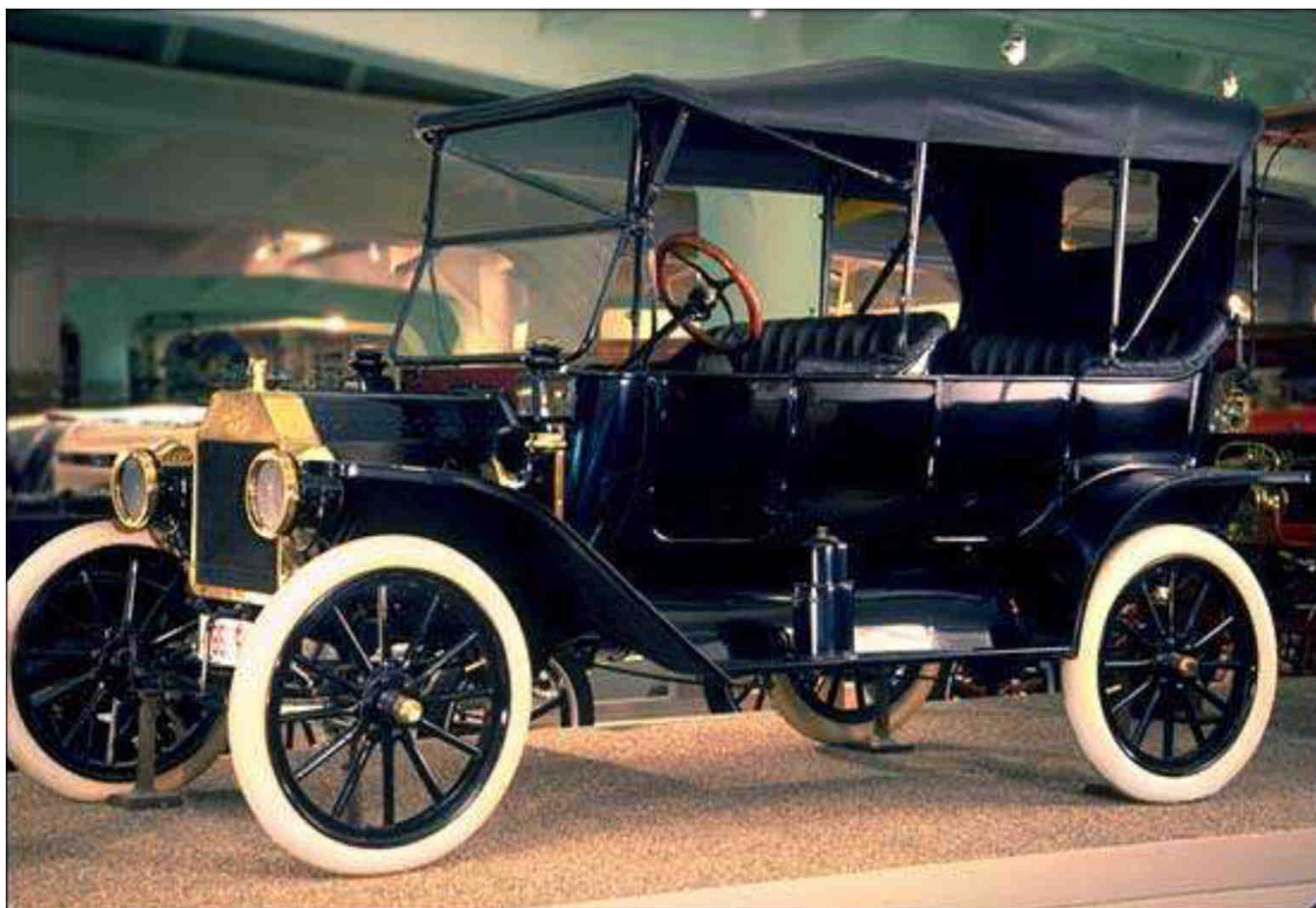
- Statistical values given by the National Greenhouse Gas Inventory
- Based on international agreement United Nations Framework Convention on Climate Change (UNFCCC)
- Mandatory IPCC methodology (Guidelines for National Greenhouse Gas Inventories etc.)
- UNFCCC parties collect data from 5 sectors:
  - ▶ Energy
  - ▶ Industrial processes
  - ▶ Agriculture
  - ▶ Land-Use, Land-Use Change and Forestry (LULUCF)
  - ▶ Waste

- The most significant category
- In central Europe > 85 % of the overall emissions of GHGs (mostly CO<sub>2</sub>)
- Combustion processes (coal, biomass, petroleum, natural gas)
- Processes joined with mining, conversion and manufacturing of fuels and energy (refineries, fugitive emissions of CH<sub>4</sub> from coal mining etc.)
- Emissions from local transport and other mobile sources (NOT international and air transport)
- Part of the fuel consumptions reported in other categories, or not taken into account:
  - ▶ non-energetic utilization of fuels for production of lubricants, asphalt etc.
  - ▶ coke as reducing agent for Fe production
  - ▶ Fuels as raw materials in chemical production, e.g. of NH<sub>3</sub>

- Coal/biomass-fired power station
- Huge flue gas flow – e.g. 200 MW unit produces:
  - ▶ Lignite:  $1.0 - 1.2 \cdot 10^6 \text{ m}^3 \text{ h}^{-1}$
  - ▶ Heavy fuel oil:  $0.5 - 0.6 \cdot 10^6 \text{ m}^3 \text{ h}^{-1}$



- Local transport vehicles
- According to the National Greenhouse Gas Inventory within ENERGY sect.
  - ▶ In 2016: 2016:  $1,32 \cdot 10^9$  (personal cars + truck + buses)

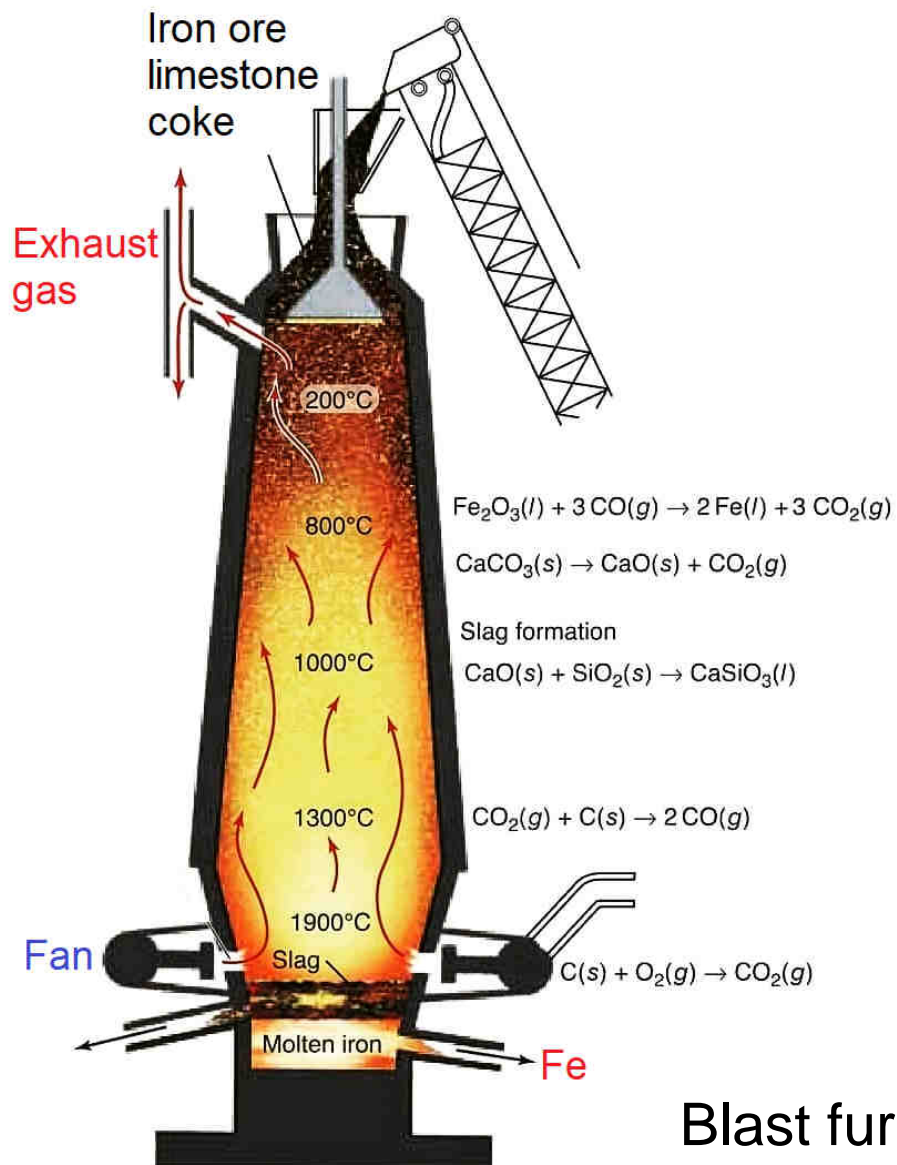


Ford model T (1908)



- Metallurgical and chemical processes
  - ▶ CO<sub>2</sub> from application of coke for reduction of iron ores to Fe
  - ▶ N<sub>2</sub>O from production of HNO<sub>3</sub>
  - ▶ CO<sub>2</sub> from production of ammonia (Haber-Bosch process) ..... etc.
- Processes of decomposition of carbonate minerals:
  - ▶ CO<sub>2</sub> from production of cement and lime
  - ▶ CO<sub>2</sub> from manufacture of glass and ceramics
  - ▶ CO<sub>2</sub> from limestone flue gas desulfurization (FGD)
- Application of F-gases:
  - ▶ HFC and PFC and (particularly in cooling and chilling processes).
  - ▶ SF<sub>6</sub> from high voltage circuits (insulating gas)
  - ▶ SF<sub>6</sub> from inert protective atmospheres (Al, Mg metallurgy)

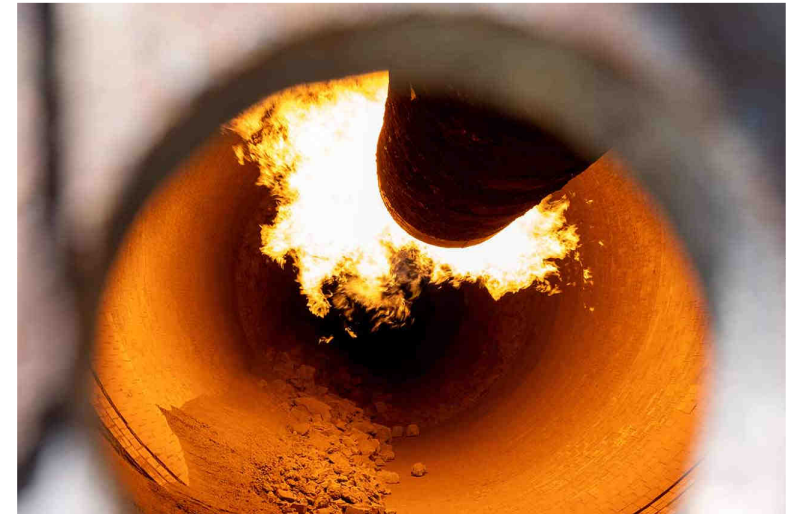
■ Examples of technologies emitting GHGs:



SF<sub>6</sub> used in wind turbines

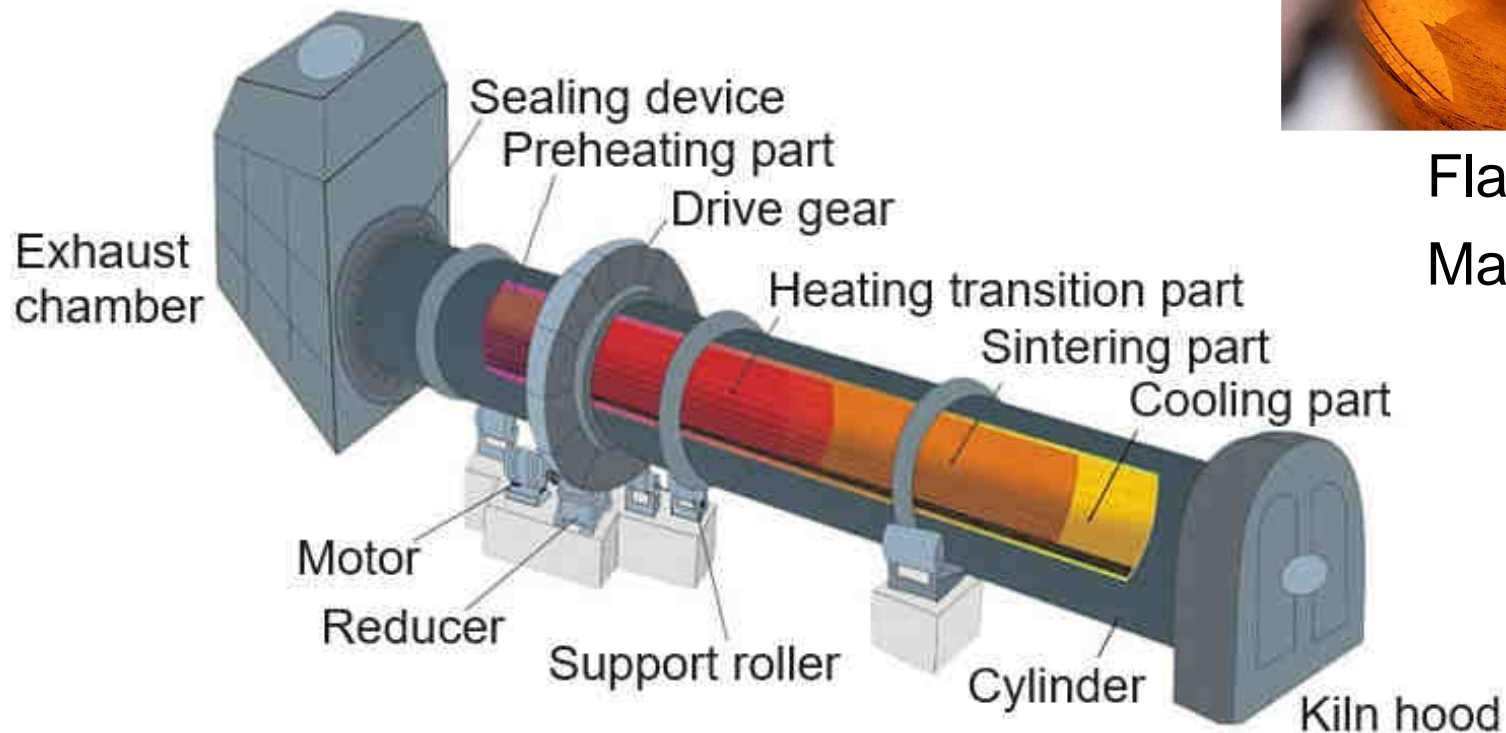
Blast furnaces emitting CO<sub>2</sub>

■ Examples of technologies emitting GHGs:



Flame: up to 2,000 °C

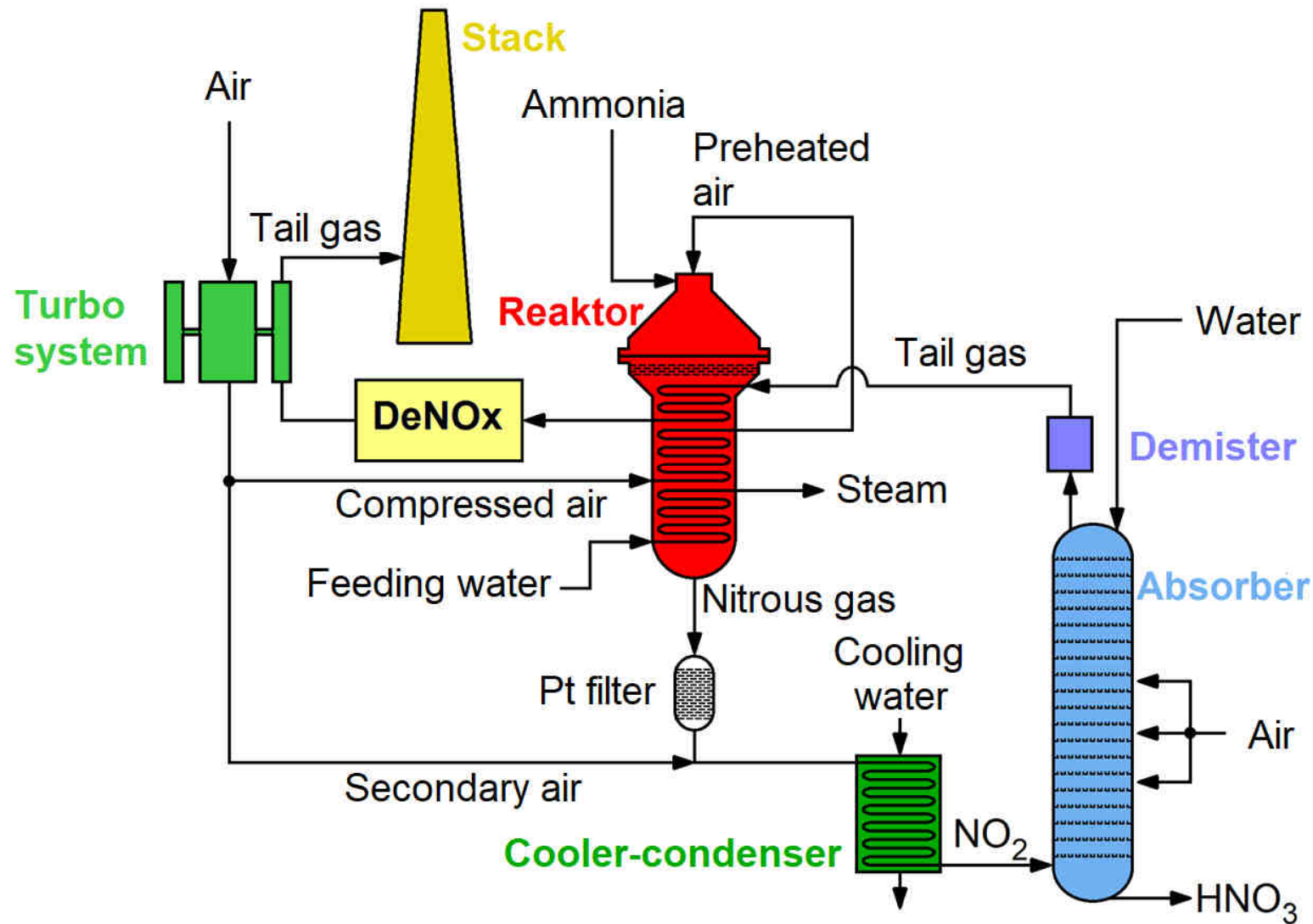
Material: up to 1,450 °C



Cement rotary kiln (combusting NG , emitting CO<sub>2</sub>)



■ Examples of technologies emitting GHGs:



Nitric acid production plant (combusting NH<sub>3</sub>, emitting N<sub>2</sub>O)

- In central Europe mostly emissions of  $\text{CH}_4$  and  $\text{N}_2\text{O}$
- Breeding of animals (anaerobic decomposition of animal manure and  $\text{CH}_4$  from enteric fermentation = digestion of vegetal aliment
  - ▶ especially breeding of bovine animals
  - ▶  $\text{CH}_4$  emissions ca. 65 - 100 mil. t/year
  - ▶ less from swine breeding
- Rice cultivation (170 mil. t/year):
  - ▶  $\text{CH}_4$  emissions ca. 170 mil. t/year
- $\text{N}_2\text{O}$  emissions from bacterial denitrification in soil



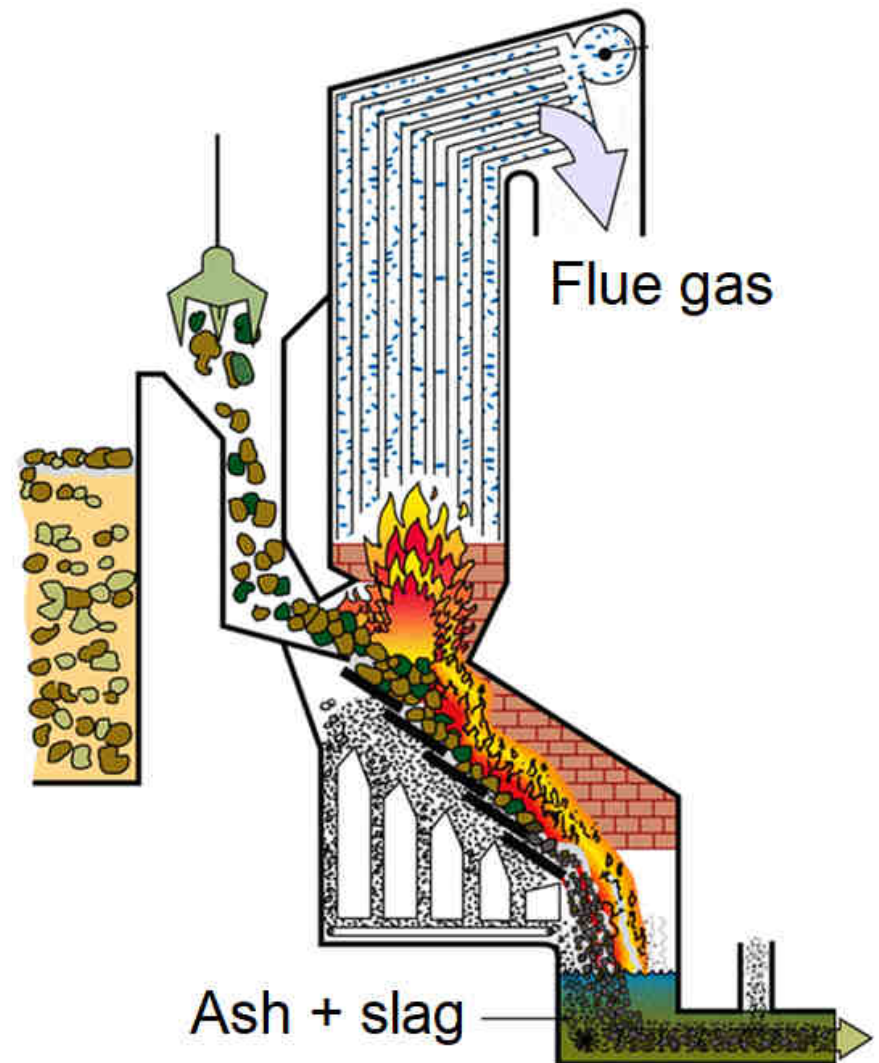
- Emissions of CO<sub>2</sub>
- Inventory based on the analysis of total quantity of wood in forests and its annual changes
- For example in the Czech Republic this sector showed higher CO<sub>2</sub> capture than it emitted
  - ▶ negative CO<sub>2</sub> balance till 2018 diminished overall emissions from other sectors
  - ▶ at present ..... forests damaged due to dry seasons ⇒ **CO<sub>2</sub> emissions**





- In central Europe mostly emissions of  $\text{CH}_4$ ,  $\text{CO}_2$ ,  $\text{N}_2\text{O}$ ;
- Treatment of municipal and industrial wastewater ( $\text{CH}_4$ ,  $\text{N}_2\text{O}$ ):
  - ▶ reported emissions of  $\text{CH}_4$  reduced by collected and energetically exploited biogas
- Municipal waste dumps ( $\text{CH}_4$ ) – landfills:
  - ▶ reported emissions of  $\text{CH}_4$  reduced by collected and energetically exploited biogas
  - ▶ 2 methods for evaluation of  $\text{CH}_4$  emissions:
    - ▶ carbon compounds converted to  $\text{CH}_4$  within the year of deposition
    - ▶ mathematic model of slower, gradual decomposition

- Waste incinerators - emissions of CO<sub>2</sub>
  - ▶ example: waste-to-energy plant



- National Inventory Report – complete data for each country
- Administrator: United Nations Framework Convention on Climate Change (UNFCCC)



- Link:

▶ [https://unfccc.int/ghg-inventories-annex-i-parties/2010 ..... 2021, 2022 etc.](https://unfccc.int/ghg-inventories-annex-i-parties/2010.....2021,2022etc)

- Complete reports sorted by years and in alphabetic order:



Party	Date of CRF original submission	Latest submitted NIR	Latest submitted CRF <sup>1</sup>	Latest submitted SEF	Status Report
Australia	15 April 2021	NIR 15 Apr 2021	CRF 15 Apr 2021	SEF-CP1-2020 15 Apr 2021  SEF-CP2-2020 15 Apr 2021	SR/AUS 10 Jun 2021
Austria	15 April 2021	NIR 15 Apr 2021	CRF 15 Apr 2021	SEF-CP1-2020 20 May 2021  SEF-CP2-2020 15 Apr 2021	SR/AUT 10 Jun 2021

1. [https://ossarchive.adm.ntu.edu.sg/2016-17/cm8001-group-30/index.html\\_p=103.html](https://ossarchive.adm.ntu.edu.sg/2016-17/cm8001-group-30/index.html_p=103.html)
2. <http://earthobservatory.nasa.gov/Features/EnergyBalance/page4.php>
3. <https://climate-woodlands.extension.org/radiative-forcing/>
4. Kiehl and Trenberth, 1997
5. <https://www.priyamstudycentre.com/2022/12/infrared-spectroscopy-ir-spectrophotometer.html>
6. <https://g.cz/galerie/dedecek-automobil-slavi/?back=/budvar-fordt/&img=2>
7. <https://www.metallics.org/pig-iron-bf.html>
8. <https://newatlas.com/energy/mingyang-myse-18x-28x/>
9. <https://www.cementplantequipment.com/products/rotary-kiln/>
10. <https://www.flukeprocessinstruments.com/en-us/service-and-support/knowledge-center/infrared-technology/temperature-monitoring-rotary-kiln-shell>
11. <https://www.ochranazvirat.cz/2020/04/14/chov-skotu/>
12. <https://farm-cs.desiguspro.com/posadka/ogorod/zlaki/ris/gde-i-kak-rastet.html>



13. <https://www.drevoastavby.cz/vse-o-drevostavbach/jak-na-drevostavbu/na-co-si-dat-pozor/5333-ohrozuje-kurovec-i-drevostavby>
14. <https://www.nazeleno.cz/energie/energetika/spalovani-odpadu-kolik-vyrobime-tepla-a-elektriny.aspx>