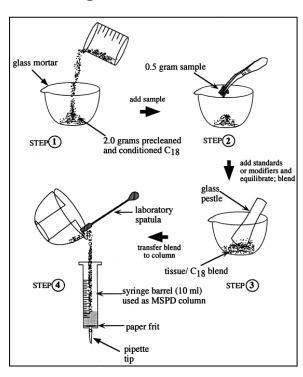


Matrix Solid Phase Dispersion (MSPD)

- blending of sample with suitable sorbent (e.g. silica gel- C_{18})
- transfer of the obtained mixture to column (syringe-like)
- (addition of other cleaning-up sorbents e.g. Florisil)
- compression of column
- elution of analytes by solvent(s)

Advantages:

- isolation and clean-up in one step
- time and solvents saving
- non-laborious



Matrix Solid Phase Dispersion (MSPD)

Blending of sample with sorbent (sample : $C_{18} = 1 : 4$)

Membrane disruption – mechanical and hydrophobic forces (lipids release)

Solid phase solvates and disperses sample components

 $(\text{non-polar} \rightarrow C_{18}, \text{polar} \rightarrow -\text{OH silica gel})$

Large contact area of phases

Matrix itself is creating "a new sorption phase"

- → affecting of the separation process
 - → analytes are eluted in fractions <u>not corresponding</u> to a simple system: analyte / pure solid phase / solvent)

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- sample extraction acetonitrile + NaCl and/or MgSO₄, + (acidification + other ...)
- shaking \rightarrow centrifugation \Rightarrow salting out of analytes to organic phase
- crude extract + MgSO₄ or sorbent PSA = Primary Secondary Amine
- shaking → centrifugation









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- removal of polar matrix components
 - organic acids, polar pigments, carbohydrates
- acidification for some analytes X degradation of others

Advantages:

- isolation and clean-up in one step
- time and solvents saving
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