

LIQUID CHROMATOGRAPHY

RECENT TECHNIQUES

HPLC – High Performance Liquid Chromatography

RRLC – Rapid Resolution Liquid Chromatography

UPLC – Ultra Performance Liquid Chromatography

UHPLC – Ultra High Pressure Liquid Chromatography

Principles of liquid chromatography

Principle generally

Separation based on different retention in chromatographic system - sorbent (stationary phase) and eluent (mobile phase)



Classification of methods

- configuration: column, thin layer ...
- separation parameters: efficiency, speed ...
- stationary phase: elution, ionex, chiral ...

Separation parameters

Resolution:

$$R = \frac{\sqrt{N}}{4} \cdot \left(\frac{\alpha - 1}{\alpha} \right) \cdot \left(\frac{1 + k'_2}{k'_2} \right)$$

OR

$$R = \frac{2(t_{R1} - t_{R2})}{w_1 + w_2}$$

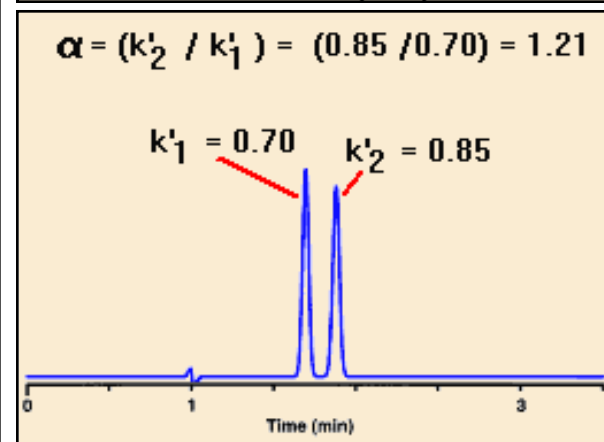
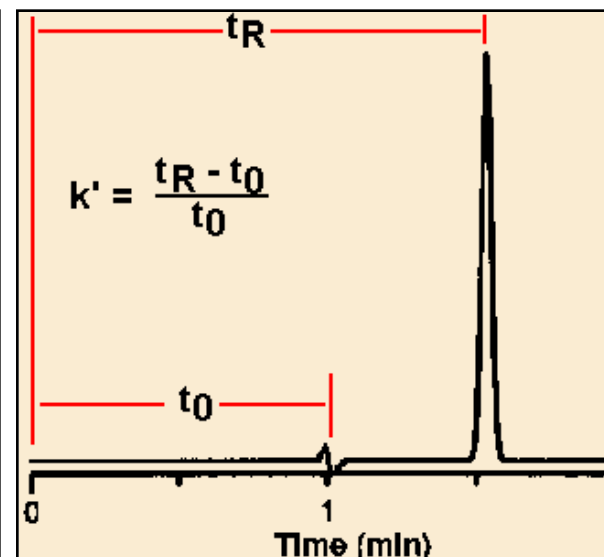
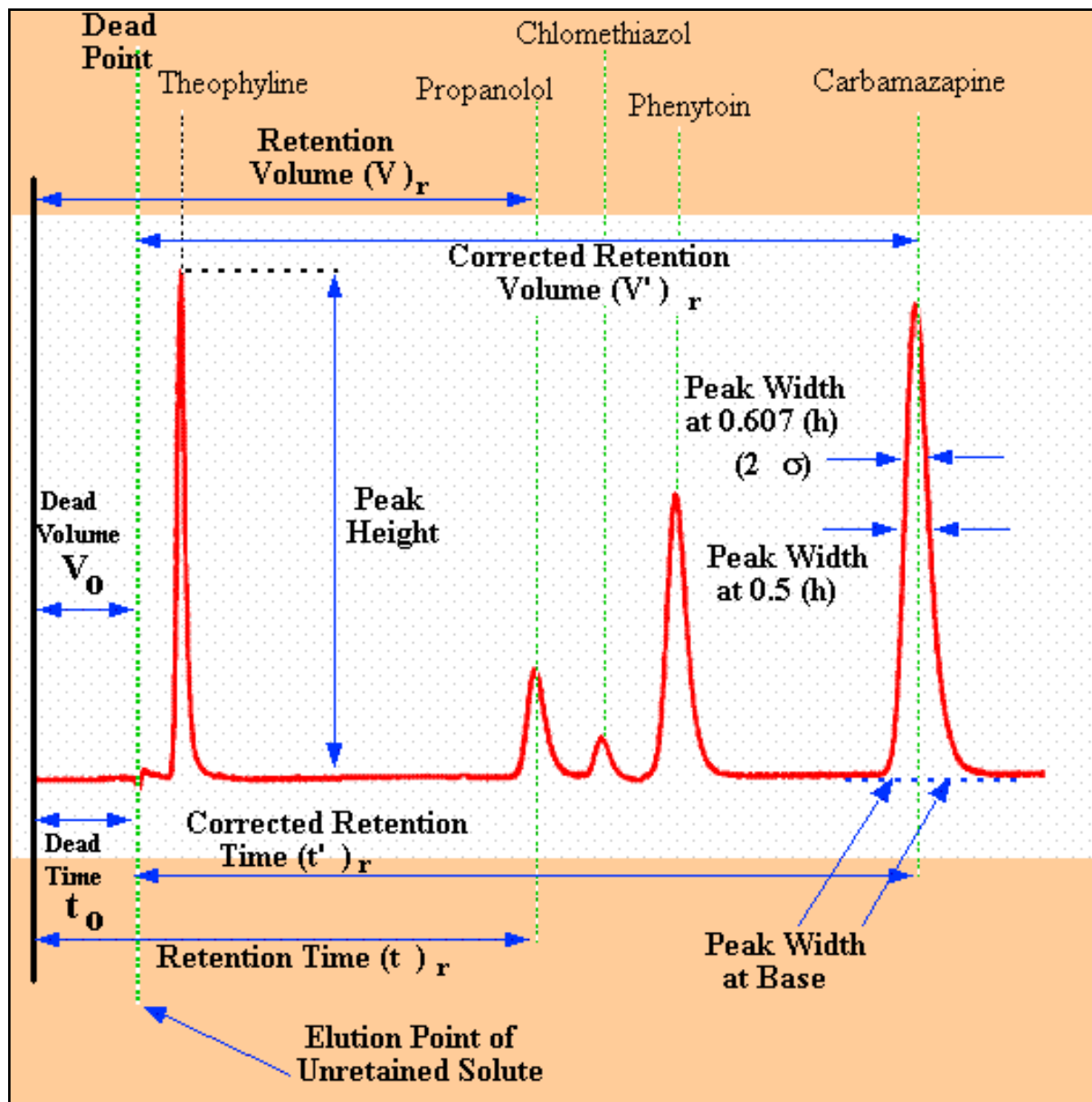
k' – capacity (retention) factor:

$$k' = \frac{n_s}{n_m} = K_D \cdot \frac{V_s}{V_m} = \frac{t_R - t_0}{t_0}$$

α – separation factor: k'_2/k'_1

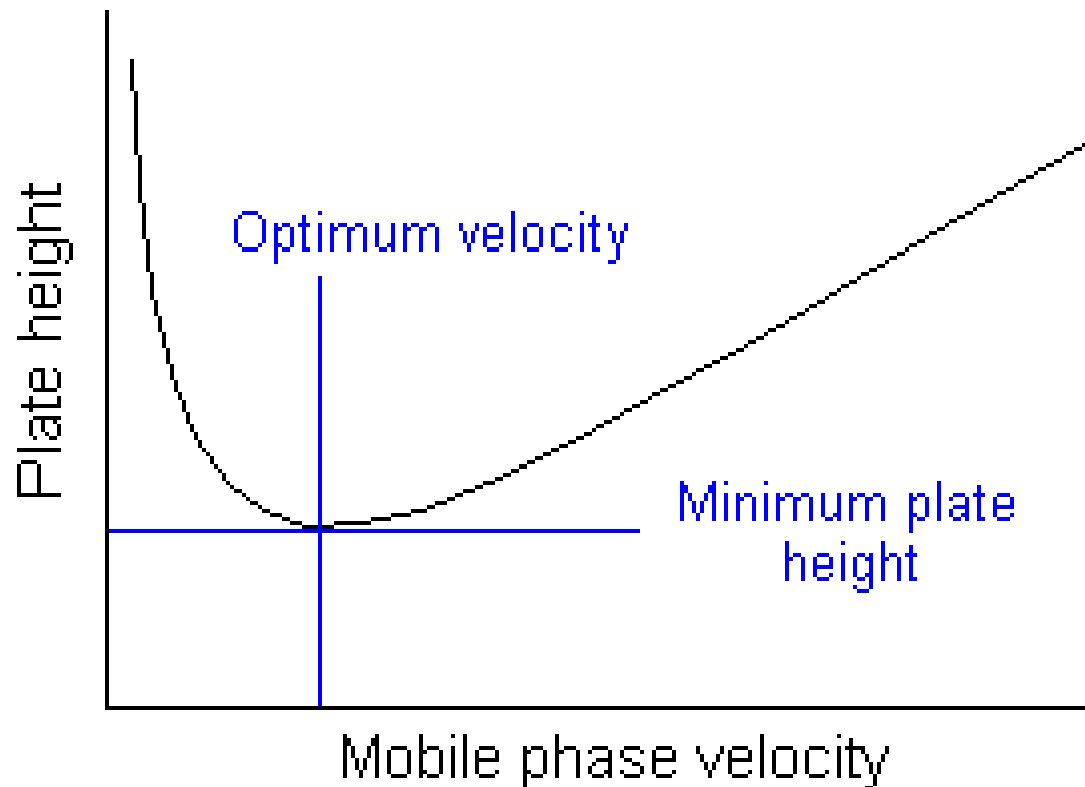
N – number of theoretical plate (efficiency)

$$N = \frac{5,55 \cdot t_R^2}{w_{1/2}^2}$$



**$R = 1,5 \Rightarrow$
baseline separation**

A typical Van Deemter plot



HETP = L/N – height equivalent of a theoretical plate

Y - HETP

X - average linear velocity

Interpretation of separation parameters

k' *~ composition of stationary and mobile phase, temperature, flow rate, column length*

- characterises retention under given conditions

- **retention factor (time of analysis)**

α *~ composition of stationary and mobile phase, temperature, flow rate, column length*

- characterises separation (comparison of k')

- **selectivity factor (analytes separation)**

N *~ stationary phase quality - particles size and sorting, mobile phase composition, flow rate, column length, temperature*

- characterises quality of elution zone - wide, shape ...

- **peak shape factor**

R *~ combined parameter*

Stationary phase – type selection (selectivity) and particles size/sorting/porosity

Elution:

- normal phase
(classic x HILIC)
- reverse phase
(classic x semipolar)
- ion pair
(reverse phase)

Ion exchange (IE)

- catex, anex

Chiral

- cyclodextrin phases

Affinity

HPLC: 3-10 μm - various
- p up to 400 bar

RRLC: 1.8 μm - porous mixed
- p up to 600 bar

UPLC: 1.7 μm - porous unified
- p up to 1000 bar

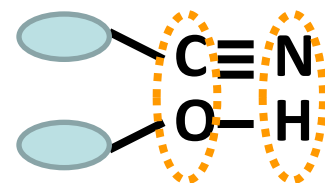
UHPLC: 1.0 μm - non-porous
- p up to 5000 bar

Interaction occurring during separation

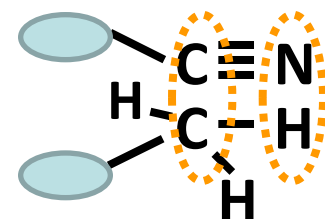
* van der Waals dispersion force – *London's induced dipole – induced dipole*



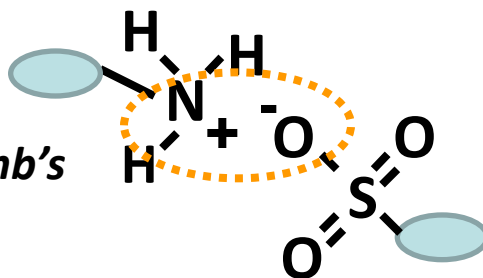
* van der Waals orientation forces – *Keesom's dipole - dipole*



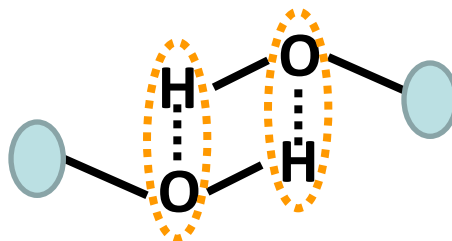
* van der Waals induction forces – *Debye's dipole – induced dipole*



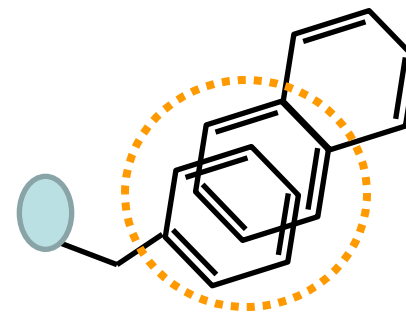
** electrostatic forces – *Coulomb's*

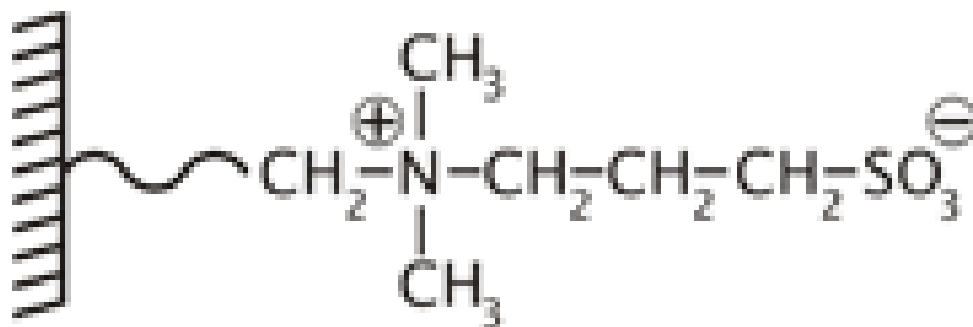
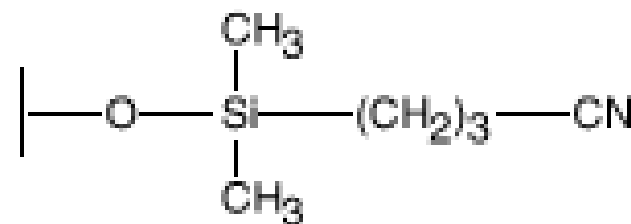
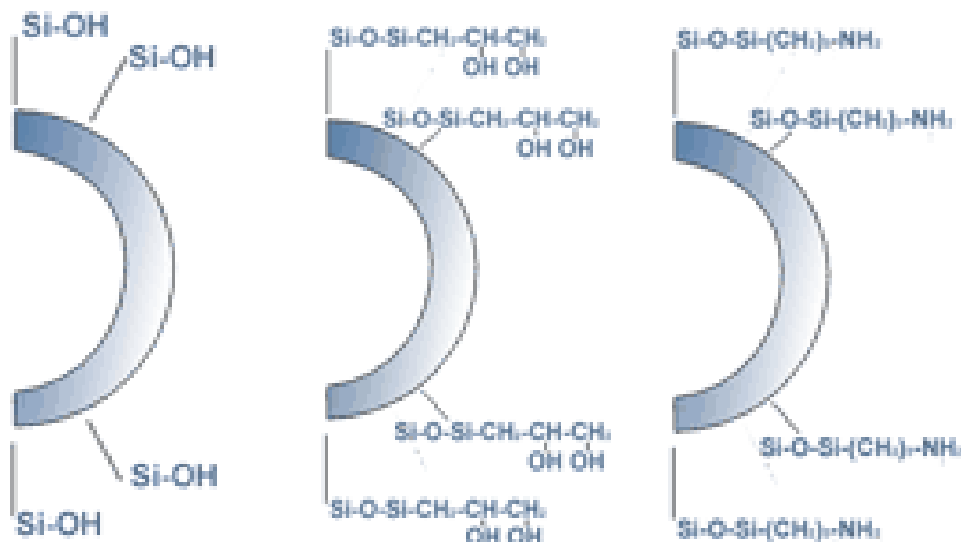


*** hydrogen bond



**** $\pi - \pi$ interaction

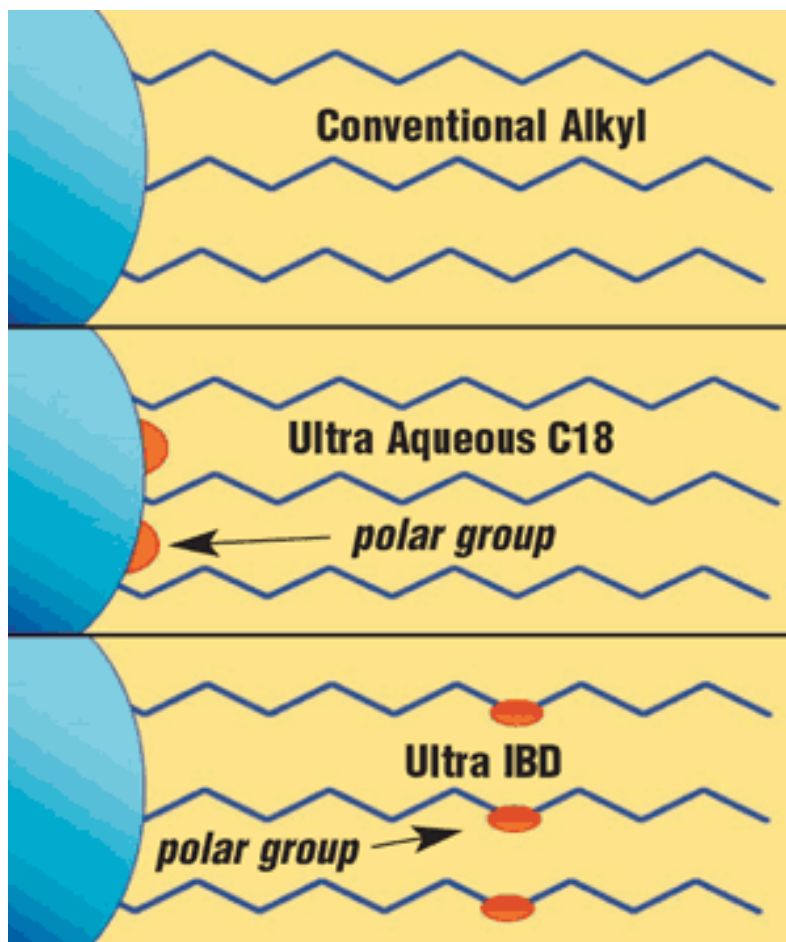


Normal phase chromatography:**Stationary phase - polar: silica gel, modified silica gel****Mobile phase - nonpolar (retention according to m.ph. comp.)****HILIC****Hydrophilic****Interaction****Liquid****Chromatography**

Reverse phase chromatography:

Stationary phase - nonpolar: modified silica gel (C18)

Mobile phase - polar (retention according to m.ph. comp.)



**ODS - octadecylated
silica gel (C18)**

Other modification: C4, C8, C30

Endcapping (Embedding):

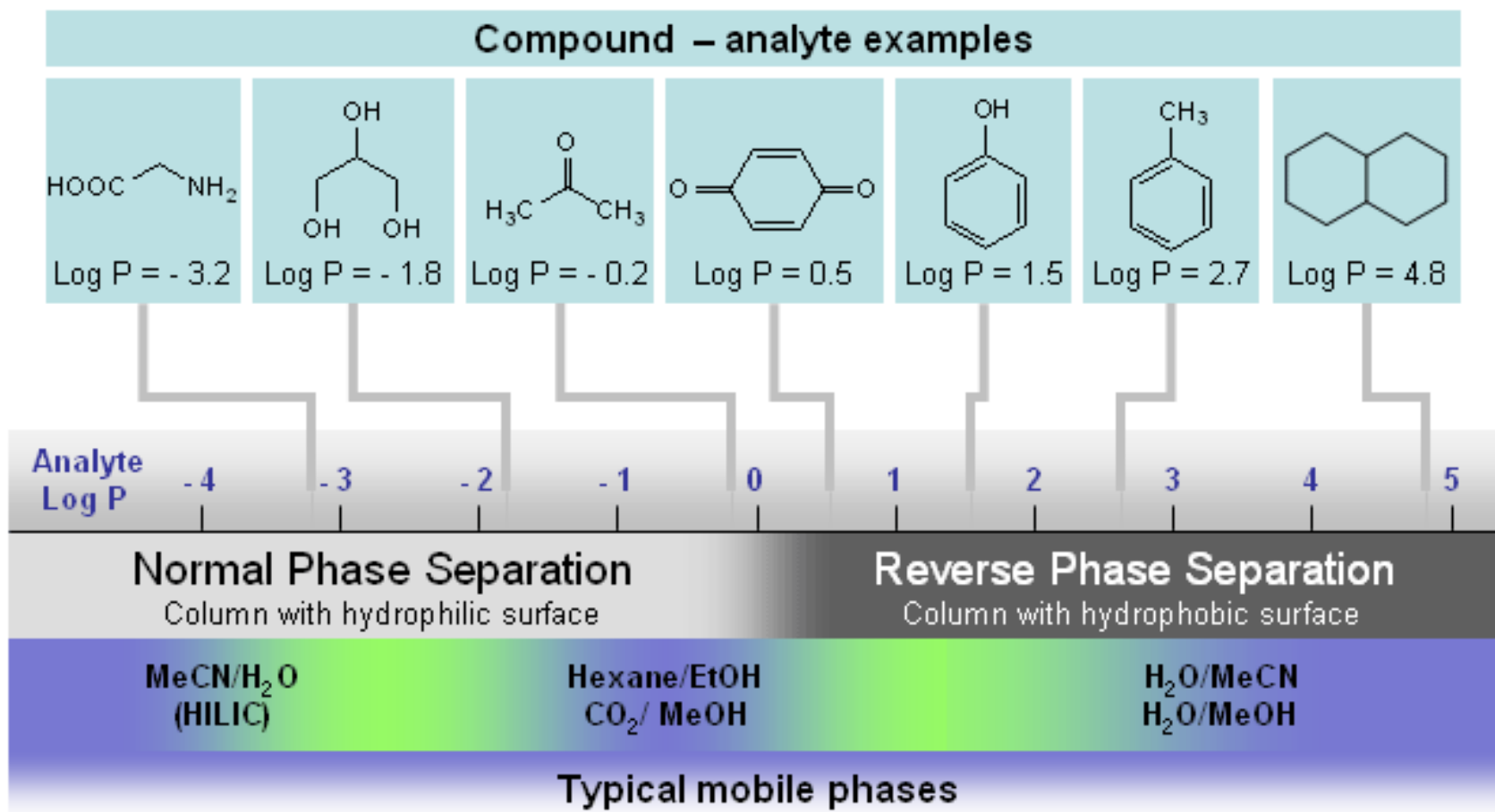
Classic - 100% nonpolar

Polar – partially polar groups

Porous x nonporous

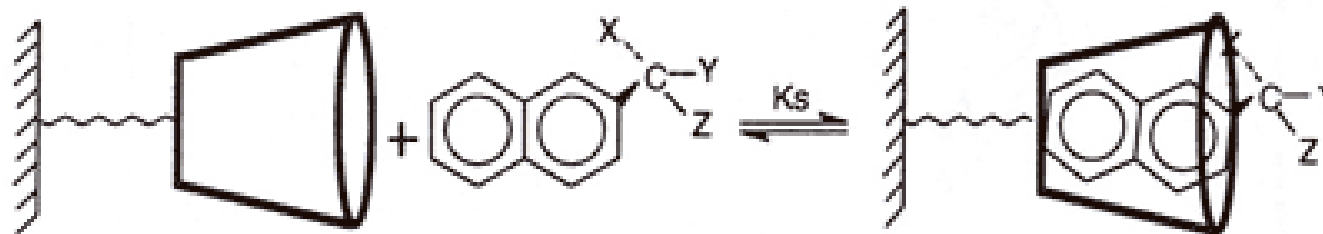
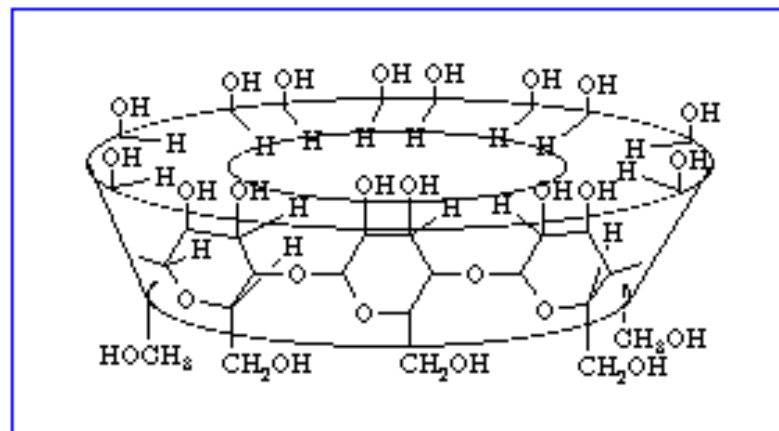
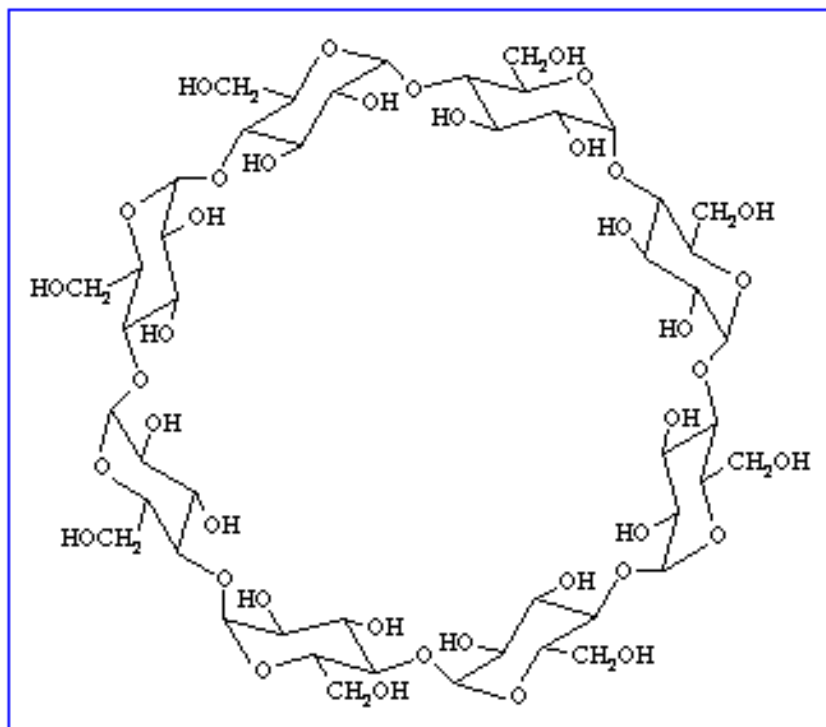
**Monomeric X polymeric
(cross-linked)**

Comparison of HILIC, NP a RP chromatography



Chiral chromatography:

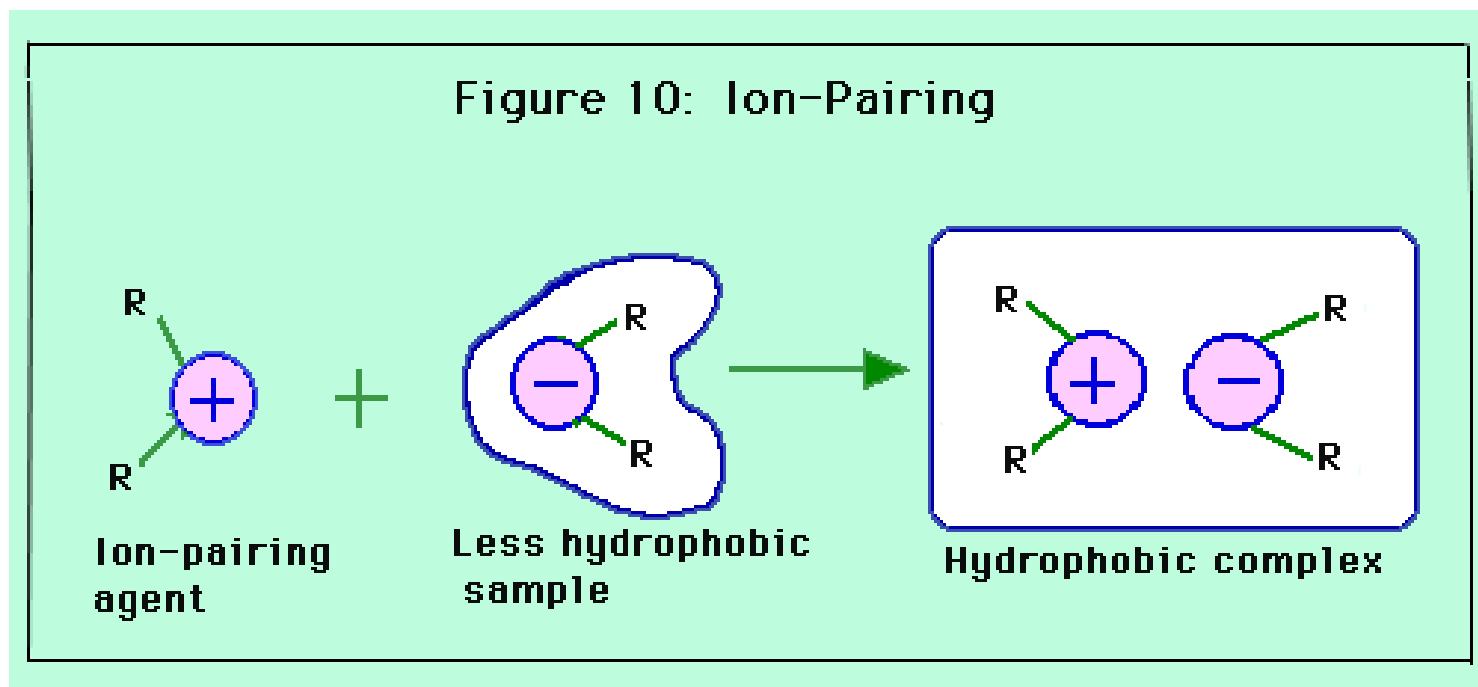
Stereoisomers (enantiomers, diastereoisomers ...)
- separation interaction with chiral phase



Ion pair chromatography:

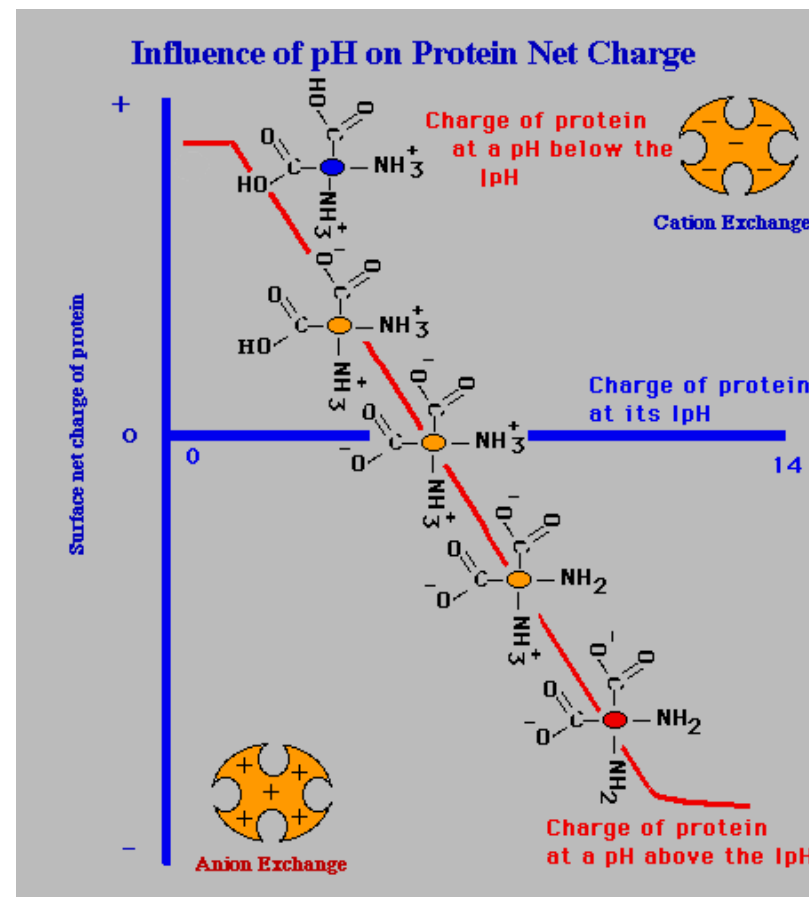
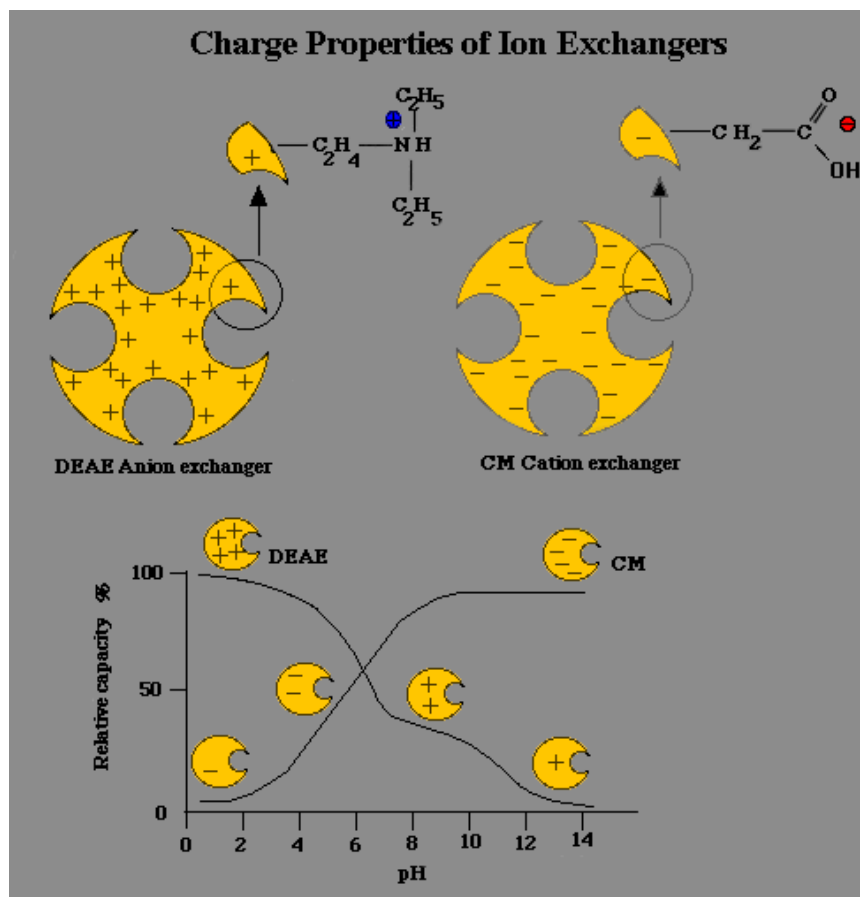
Polar analyte + polar agent (ion pair agent)

**⇒ nonpolar complex with retention on nonpolar st.ph.
- realisation on reverse phase**



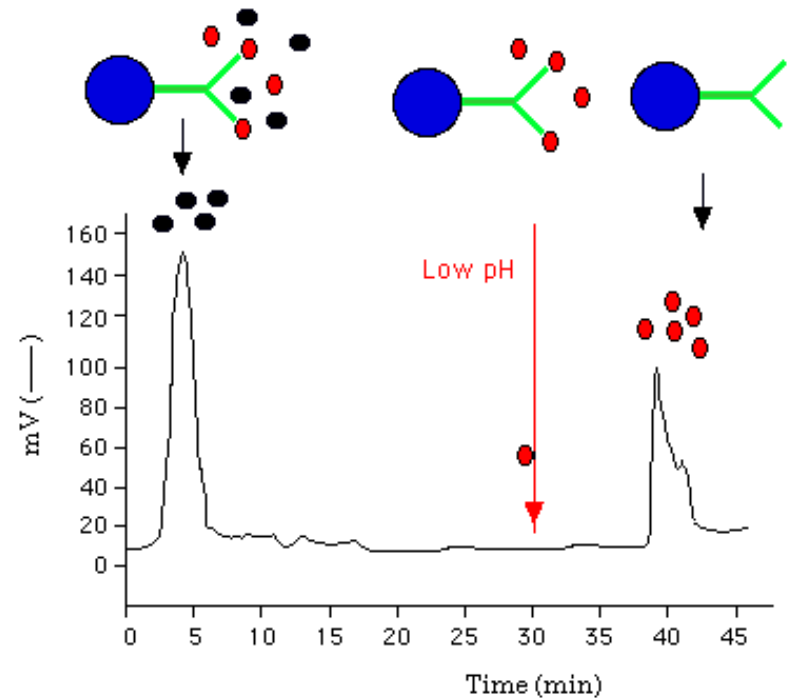
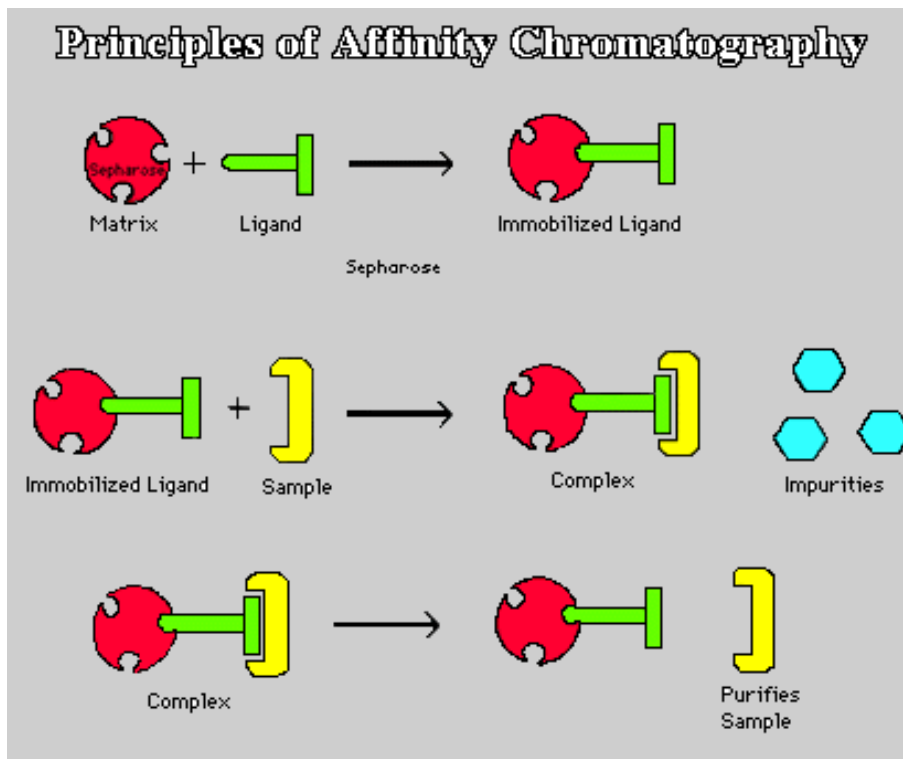
Ion exchange chromatography:

**Stationary phase - embedding \oplus or \ominus ; counter ions are mobile
(retention according to m.ph. composition)**



Affinity chromatography:

**Stationary phase – immobilisation of ligand (biologically active compounds);
sample is bonded to complex and released
by the change of mobile phase)**



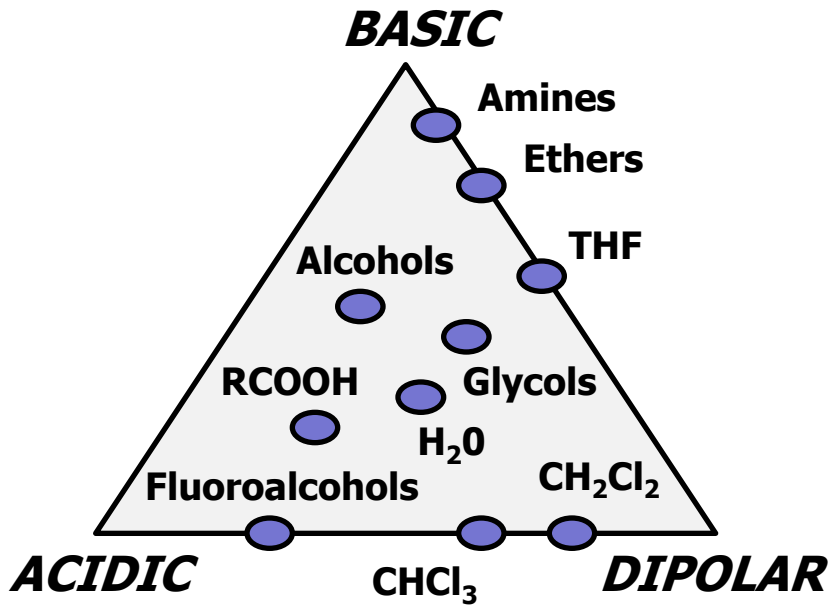
Mobile phase - polarity, ion strength, elution technique

SELECTION:

Sample solubility

Compatibility with stationary phase and system

Elution: isocratic ...X... gradient



Solvent Miscibility Table

Dielectric Constant	UV (nm) Cutoff	Refractive Index	Viscosity (cP)	Boiling Point(°C)	Solvent
1.94	210	1.3914	0.50	98	iso-Octane
1.88	195	1.3749	0.31	69	n-Hexane
1.92	200	1.3876	0.41	98	n-Heptane
4.33	218	1.3524	0.24	34	di-Ethyl ether
2.02	200	1.4262	1.00	80	Cyclohexane
6.02	256	1.3724	0.45	76	Ethyl acetate
2.38	284	1.4969	0.59	110	Toluene
4.81	245	1.4458	0.57	60	Chloroform
7.58	212	1.4072	0.55	65	Tetrahydrofuran
2.27	278	1.5011	0.65	80	Benzene
20.70	330	1.3587	0.36	56	Acetone
8.93	233	1.4241	0.44	40	Dichloromethane
2.25	215	1.4224	1.37	101	Dioxane
20.33	210	1.3856	2.30	98	n-Propanol
25.80	210	1.3610	1.20	78	Ethanol
36.70	268	1.4305	0.92	153	Dimethylformamide
37.50	190	1.3411	0.38	82	Acetonitrile
6.30	230	1.3720	1.26	118	Acetic Acid
4.70	260	1.4830	2.24	189	Dimethyl Sulfoxide
32.70	205	1.3284	0.55	65	Methanol
81.10	210	1.3330	1.00	100	Water

Mobile phase – selection according to elution strength

Polarity (Snyder's polarity index) P'

Eluotropic value – elution strength (stationary phase type, primary for NP) - ϵ^0

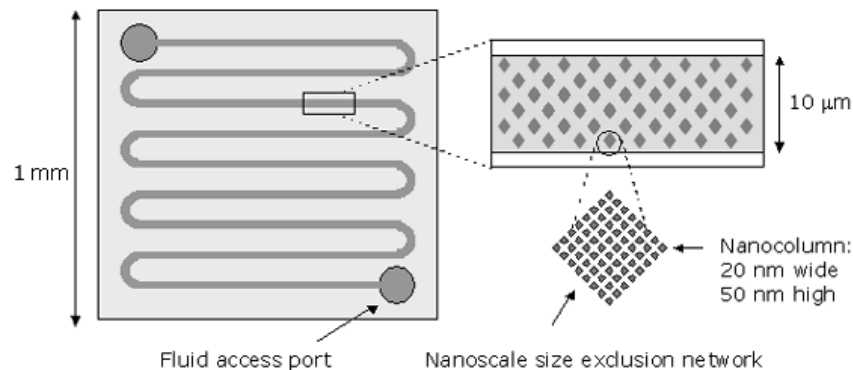
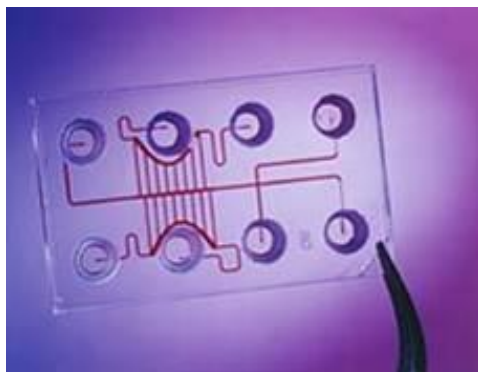
Solubility parameters ... δ

Solvent	P'	ϵ^0 (silica gel)	ϵ^0 (C18)	δ
Hexane	0.1	0	-	7.3
Acetone	5.1	0.53	8.8	9.6
Acetonitrile	5.8	0.52	3.1	12.7
Methanol	5.1	0.7	1.0	14.5
H₂O	10.2	-	-	23.5

Typical column and sorbents - technical parameters

<i>Column dimensions</i>	<i>Particles</i>	<i>Flow rate</i>	<i>Chromatography</i>
10-25 cm x 3-4.6 mm	3-10 μm	0.1-2 mL/min	Conventional (classic)
0.5-7.5 cm x 3-4.6 mm	3-10 μm	0.1-2 mL/min	fast
5-25 cm x \pm 2 mm	3-10 μm	0.05-1 mL/min	narrow x micro-bore 2
15-100 cm x \pm 1 mm	3-10 μm	30-60 $\mu\text{L}/\text{min}$	narrow x micro-bore 1
10-200 cm x 0.2-0.5 mm	1-5 μm	1-10 $\mu\text{L}/\text{min}$	capillary (packed)
100-200 cm x 0.04-0.08	-----	< 1 $\mu\text{L}/\text{min}$	capillary (open tubular)
0.5-2 cm x 2 mm	1.8 μm	0.5-2.5 mL/min	RRLC
5-15 cm x 2 mm	1.7 μm	0.1-1.0 mL/min	UPLC
5-15 cm x < 1-2 mm	1.0 μm	< 0.1-0.5 mL/min	UHPLC
Grooves on a chip	2.5-10 μm	< 1 $\mu\text{L}/\text{min}$	nano-LC (Lab on Chip)

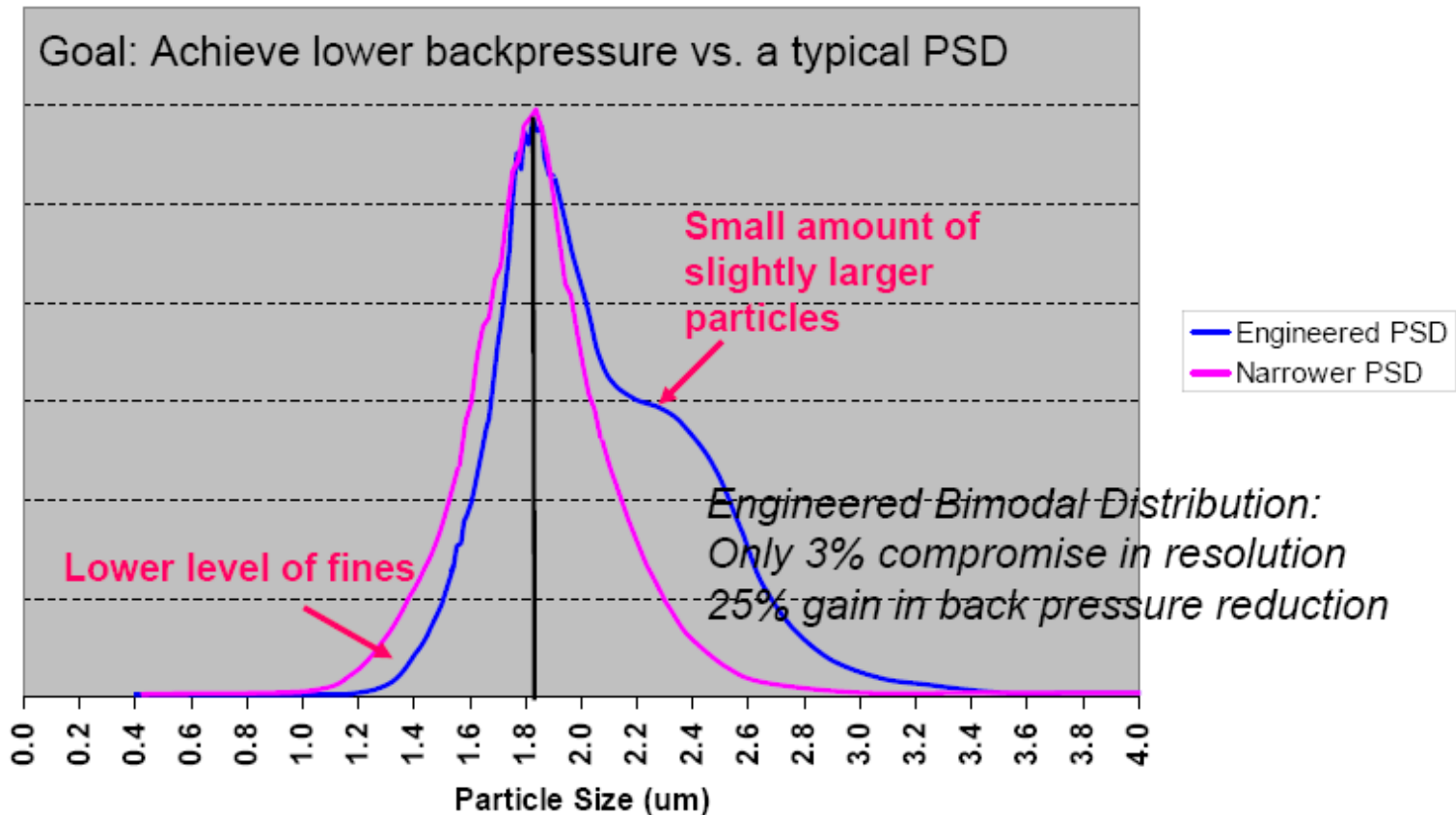
**Lab
on
Chip**



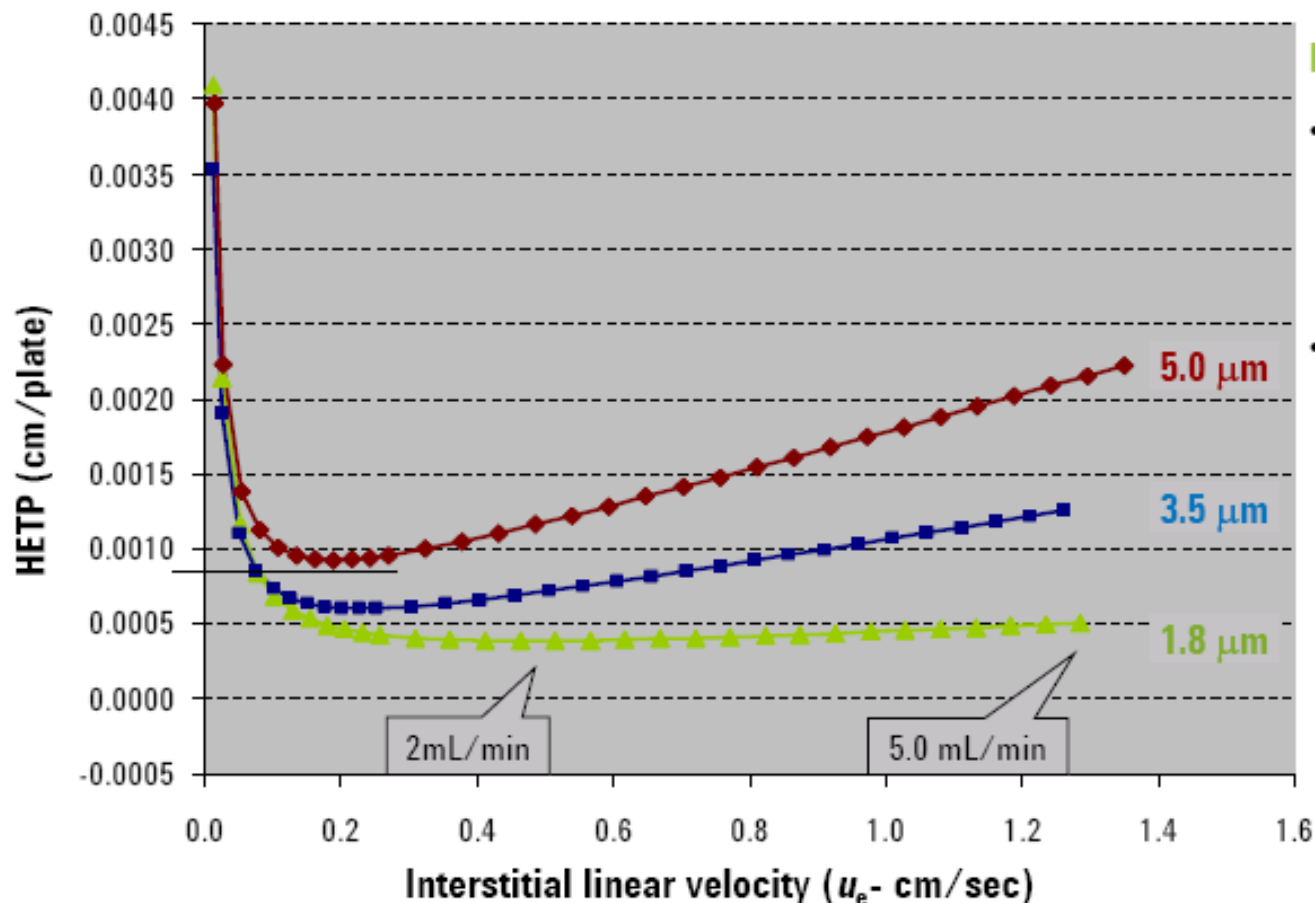
Sorbent sorting - back pressure effect



- higher permeability \Leftrightarrow porosity
- can be higher than sorbent particles
- lower permeability \Leftrightarrow porosity
- must be less than sorbent particles \Rightarrow corresponding back pressure



Small particles – sorbents application

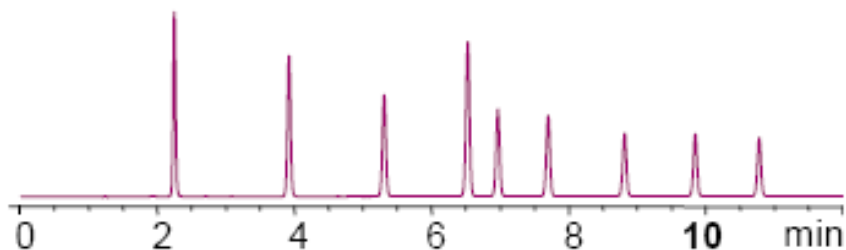


Increased Speed

- Run short columns with small particle size at high linear velocities
- Speed gains: 5-10x

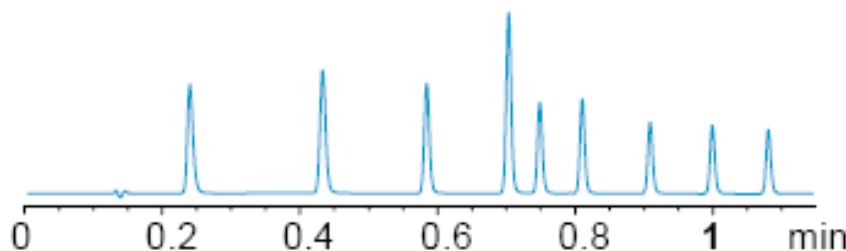
Application of rapid resolution (RRLC)

4.6 x 150mm, 5µm
1.20ml/min, 40°C
Analysis Time = 11min



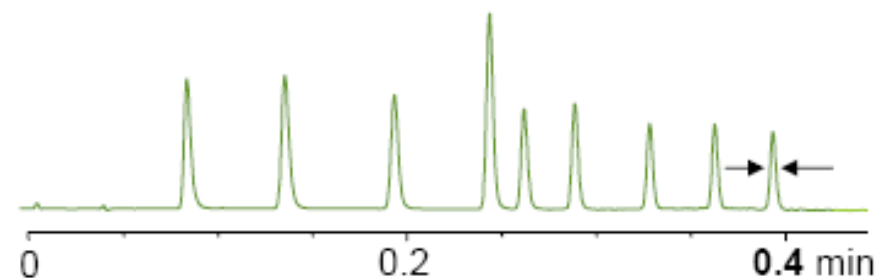
HPLC, 40°C
PW = 3.4sec

2.1mm x 50mm 1.8µm
1.00ml/min, 40°C
Analysis Time= 1.1min



RRLC, 40°C
10x faster
PW = 0.5 sec

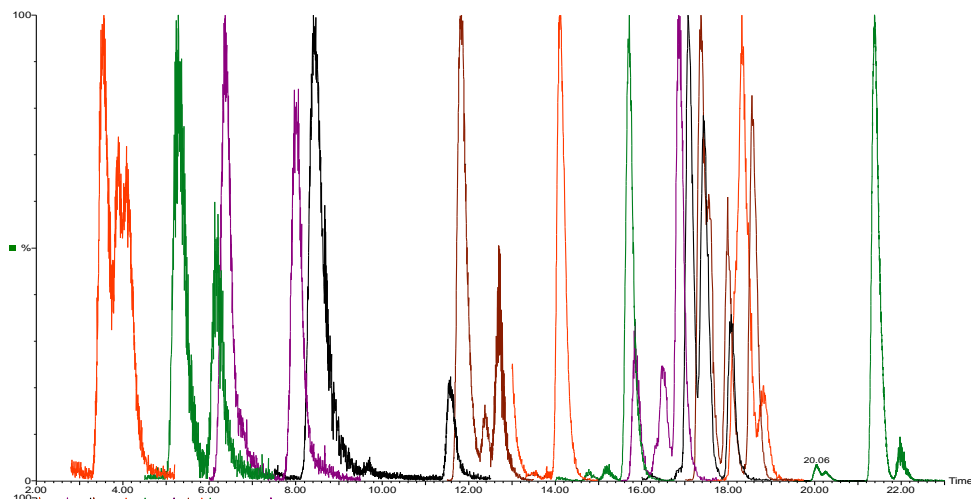
2.1mm x 50mm 1.8µm
2.40ml/min, 95°C
Analysis Time: 0.4min



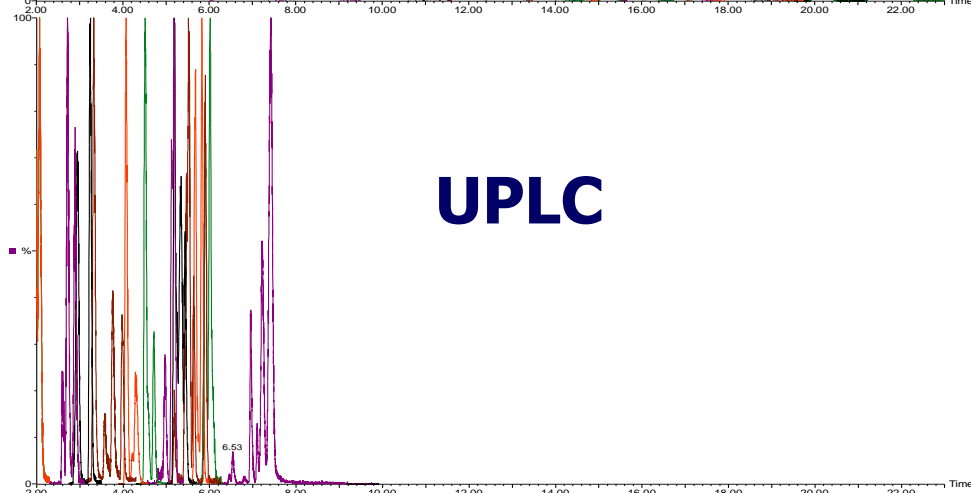
RRLC, 95°C
27x faster
PW = 197msec

150mm > 50mm: } 3x
1.2ml/min on 4.6 > 2.4ml/min on 2.1: } 10x } 3 x 10 = 30x

Application of UPLC - multiresidue pesticide analysis



HPLC



UPLC

**Flow rate: 0.3 mL/min
(in both systems)**

UPLC 10 min

HPLC 50 min