



(2,85 t)
c.kmpta

$$m_S = 1000 \text{ kg}$$

$$m_{O_2} = n_S = \frac{m_S}{M_S} = \frac{1000 \text{ kg}}{32 \text{ kg/kmol}}$$

$$m_{O_2} = m_{O_2} \cdot M_{O_2} = \frac{1000}{32} \cdot 32 = 1000 \text{ kg}$$

$$\frac{m_{O_2}}{m_{N_2}} = \frac{23,3}{76,7} \Rightarrow m_{N_2} = \frac{76,7}{23,3} \cdot 1000 = 3439,5 \text{ kg}$$

$$m_{SO_2} = m_{SO_2} \cdot M_{SO_2} = \frac{1000}{32} \cdot 64 = 2000 \text{ kg}$$

$$m_{SO_2} = n_S$$

$$Q = m_p (h_{para}^{120^\circ C} - h_{voda}^{25^\circ C})$$

\uparrow 2706 kJ/kg \uparrow 104,8 kJ/kg

$$m_p = \frac{Q}{2601,2 \text{ kJ/kg}}$$

Ref. stav 25°C
látky ve formě prvků

$$H^{in} = H^{out} + Q$$

$$H^{in} = 0 \text{ (na vstupě jsou jen prvky v ref. teplotě)}$$

$$H^{out} = m_{SO_2}^{out} h_{SO_2}^{out} + m_{N_2}^{out} h_{N_2}^{out}$$

$$h_{SO_2}^{out} = \Delta_f h_{SO_2} + \langle c_{p,SO_2} \rangle \cdot (400 - 25) = -4638 + 0,720 \cdot 375 = -4368 \text{ kJ/kg}$$

$$h_{N_2}^{out} = \langle c_{p,N_2} \rangle \cdot (400 - 25) = 1,054 \cdot 375 = 395,25 \text{ kJ/kg}$$

$$Q = H^{in} - H^{out} = 0 - \{-2000 \cdot 4368 + 3439,5 \cdot 395,25\} = 7,377 \cdot 10^6 \text{ kJ}$$

$$m_p = \frac{7,377 \cdot 10^6}{2601,2} = 2836 \text{ kg páry}$$