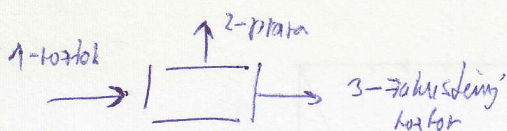


U2-2

57 kg/h 54% m A 12 kg/h voda  
58% m voda



A-	0,44	—	$w_{A3}$
B-voda	0,56	1	$w_{B3}$
	57	12	$m_3$
	kg/h	kg/h	

$$0,44 \cdot 57 = w_{A3} \cdot m_3$$

$$57 + 12 = m_3 + 12$$

$$m_3 = 55 \text{ kg}$$

$$w_{A3} = 0,557$$

U2-4

Cl 1,60% O<sub>2</sub>

3,60% O<sub>2</sub>

5 min + 33 s → 20 kg O<sub>2</sub>

$\dot{V} = ?$  priv. pouden

$$M = \frac{m}{n}$$

A-Cl	0,016	0,984	—	0,964
B-O <sub>2</sub>		0,016	1	0,036
	$n_1$	$n_2$		$n_3$

$$M(O_2) = 32 \frac{\text{kg}}{\text{kmol}}$$

$$= 20 \frac{\text{kg}}{\text{kmol}}$$

$$n_2 = \frac{m_2}{M} = \frac{20 \text{ kg}}{32 \text{ kg/kmol}} = 0,625 \text{ kmol}$$

$$n_2 = 625 \text{ mol}$$

$$0,984 n_1 = 0,964 n_3$$

$$n_1 + 625 = n_3$$

$$0,984 n_1 = 0,964 n_1 + 602,5$$

$$0,02 n_1 = 602,5$$

$$n_1 = 30125 \text{ mol za } 5 \text{ min } 33 \text{ s}$$

$$n_1 = 90,465 \text{ mol/s}$$

$$pV = nRT$$

$$V = n \frac{RT}{p} = 90,465 \cdot \frac{8,314 \cdot 293,15}{0,1 \cdot 10^5} = 2,204 \frac{\text{m}^3}{\text{s}}$$