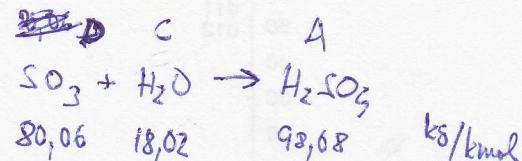
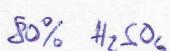
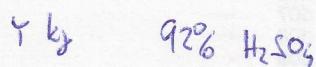
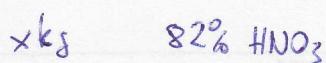
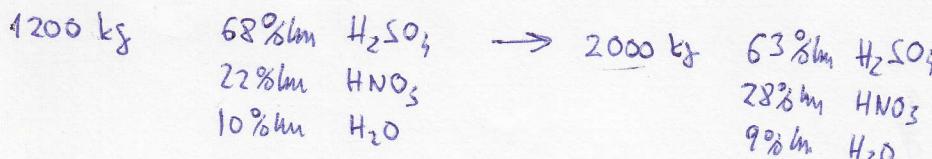


Z-11



	①	②	③	④	⑤
A H_2SO_4	0,68	—	0,92	0,80	0,63
B HNO_3	0,22	0,82	—	—	0,28
C H_2O	0,10	0,18	0,08	—	0,09
D SO_3	—	—	—	0,20	—
	1200 kg	m_2	m_3	m_4	2000 kg
	2	3	4		
	↓	↓	↓		
1-sous				5	

3+1 neznáme

4 bilancová rovnice

$$(A) \cancel{1200 \cdot 0,68 + 0,92 m_3 + 0,80 m_4 + f M_A} = 2000 \cdot 0,63 \quad (x)$$

$$(B) 1200 \cdot 0,22 + 0,82 m_2 + = 2000 \cdot 0,28$$

$$(C) 1200 \cdot 0,1 + 0,18 m_2 + 0,08 m_3 = 2000 \cdot 0,09 + f M_C$$

$$(D) 0,2 m_4 = f M_D$$

$$\Sigma 1200 + m_2 + m_3 + m_4 + (M_A + f) = 2000 + (M_B + M_C) + f$$

$$0,82 m_2 = 296 \Rightarrow m_2 = 360,98 \text{ kg (HNO}_3)$$

$$\cancel{1200 \cdot 0,68 + 0,92 m_3 + 0,80 m_4 + f \cdot 18,02} = 120,08 + f \cdot 18,02 \quad (120,08 +)$$

$$0,2 m_4 = f \cdot 80,06 \quad (10,2 +)$$

$$\underline{m_3 + m_4 = 439,02}$$

$$\underline{225,25 f + 400,3 f} = 62,25 + 439,02$$

$$625,55 f = 501,25 \Rightarrow f = 0,8014 \text{ kmol}$$

$$m_4 = \underline{320,8 \text{ kg (oleum)}}$$

$$m_3 = \underline{118,2 \text{ kg (H}_2\text{SO}_4)}$$